## Supplementary information file - Performance of nanocrystalline Ni<sub>3</sub>N as a negative electrode for sodium-ion batteries (Li, Hasan, Hector and Owen)



**Fig. S1** Voltage versus specific capacity (top) and reduction specific capacity versus cycle number (bottom) of Ni<sub>3</sub>N-lithium half cells cycled between 3 V and 1 mV at different current rates for 30 cycles (5C = 2115, 3C = 1269 and 1C = 423 mAg<sup>-1</sup>). The counter electrode was lithium foil and the electrolyte was 1 M LiPF<sub>6</sub> in EC:DEC (1:1).

Parameter (both patterns	Data from fit in Fig. 1(Ni <sub>3</sub> N	Data from fit in Fig. 2 (Ni <sub>3</sub> N	
refined in P6 <sub>3</sub> 22 with Ni at	obtained by ammonolysis of	obtained by ammonolysis of	
$^{1}/_{3}$ ,0,0 and N at $^{1}/_{3}$ , $^{2}/_{3}$ , $^{1}/_{4}$ )	[Ni(NH <sub>3</sub> ) <sub>6</sub> ].2NO <sub>3</sub> )	[Ni(EDA) <sub>3</sub> ].2NO <sub>3</sub> )	
a / Å	4.62231(14)	4.5937(4)	
c / Å	4.30630(13)	4.3258(7)	
$R_{wp}$ / %	13.0	13.5	
$R_p$ / %	9.1	10.3	
Ni $U_{iso} \times 100$ / ${\rm \AA}^2$	0.23(7)	2.93(15)	
$N~U_{iso} \times 100$ / ${\rm \AA}^2$	1.00(-)	7.2(15)	

Table S1	Refined	parameters	from	Rietveld	refinements	of Ni <sub>3</sub> N
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