Supporting information for :

Ruthenium Complex Dye with Designed Ligand Capable of Chelating Triiodide Anion for Dye-Sensitized Solar Cells

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Fig. S1 13 C NMR spectrum of **RuAS**.



Fig. S2 HSQC spectrum of RuAS.



Fig. S3 UV-vis absorption spectra of RuAS and Ru2A in ACN/t-butanol (1:1 by volume) at a concentration of 50 μ M.



Fig. S4 ¹H NMR spectra of bsacbpy and its mixtures with equivalent concentration (0.01M) of LiI₃ and LiI, respectively. Note: The proton peak at 10.7 ppm contributed by the amide groups of bsacbpy became sharper after chelating I_3^- , similar to the dimer of RuAS shown in Fig. 2.



Fig. S5 Three dimensional configurations of **RuAS** and I_3^- molecules drawn by ChemBio3D. (The blue, green, grey, purple, red and white color is nitrogen, ruthenium, carbon, iodide, oxygen and hydrogen.)



Fig. S6 Photocurrent-voltage plots of DSSC based on **Ru2A** with various I₂

concentrations of liquid electrolyte under AM 1.5 full sunlight.



(b)



Fig. S7 IPCE spectra of the DSSCs with (a) RuAS and (b) Ru2A containing various

I₂ concentrations in the liquid electrolyte.

Table S1 Photovoltaic properties of DSSC with Ru2A containing various I_2

[I ₂] (M)	PCE (%)	Jsc (mAcm ⁻²)	Voc (V)	ff
0	5.19	10.54	0.738	0.67
0.01	4.92	13.62	0.703	0.51
0.03	5.53	15.59	0.692	0.51
0.05	5.90	15.20	0.680	0.57
0.10	5.90	14.30	0.659	0.63
0.20	4.37	10.30	0.639	0.66

concentrations in the liquid electrolyte under 100 mWcm⁻² illumination.