

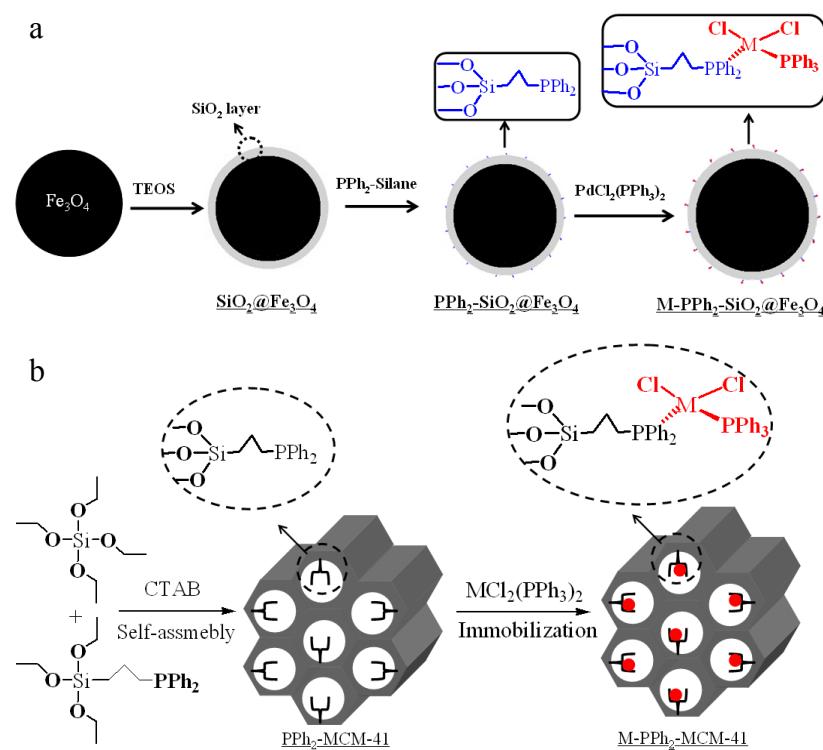
## ***Supporting Information***

# **Highly Active, Durable and Recyclable Ordered Mesoporous Magnetic Organometal Catalysts Promoted Organic Reactions in Water**

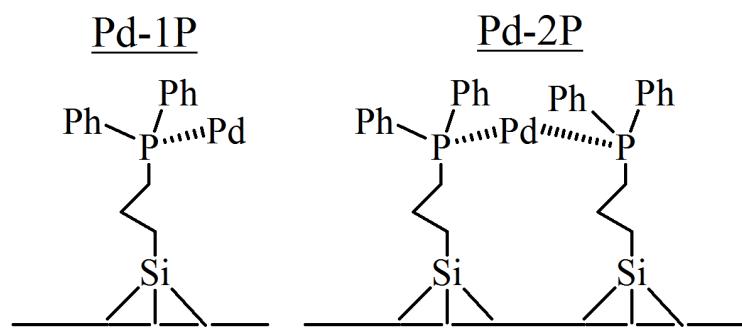
Fang Zhang,<sup>a,b</sup> Mingzhen Chen,<sup>a</sup> Xiaotao Wu,<sup>a</sup> Wei Wang<sup>b,\*</sup> and Hexing Li<sup>a,\*</sup>

<sup>a</sup> *The Education Ministry Key Lab of Resource Chemistry and Shanghai Key Lab of Rare Earth Functional Materials, Shanghai Normal University, China*

<sup>b</sup> *Department of Chemistry and Chemical Biology, University of New Mexico, Albuquerque, NM 87131-000, USA*



Scheme S1 Schematic illustration of the synthetic procedure of (a)  $\text{M-PPh}_2\text{-SiO}_2\text{@Fe}_3\text{O}_4$  and (b)  $\text{M-PPh}_2\text{-MCM-41}$  catalysts ( $\text{M} = \text{Pd}^{2+}$  or  $\text{Rh}^+$ ).



Scheme S2 Dependence of the coordination model on the P/M ratio of different catalysts.

Table S1 Elemental analysis and structural parameters of different catalysts

Sample	Pd Loading (mmmol/g)	P content ( mmmol/g)	Particle Size (nm)	S <sub>BET</sub> (m <sup>2</sup> /g)	D <sub>P</sub> (nm)	V <sub>P</sub> (cm <sup>3</sup> /g)
Pd-PPh <sub>2</sub> -SiO <sub>2</sub> @Fe <sub>3</sub> O <sub>4</sub>	0.0109	0.110	440	35	1.0	0.030
PPh <sub>2</sub> -MCM-41@SiO <sub>2</sub> @Fe <sub>3</sub> O <sub>4</sub>	/	0.380	/	268	2.9	0.43
Pd-PPh <sub>2</sub> -MCM-41@SiO <sub>2</sub> @Fe <sub>3</sub> O <sub>4</sub>	0.0723	0.503	520	240	2.8	0.37
Pd-PPh <sub>2</sub> -MCM-41	0.161	0.516	10~1500	551	2.5	0.60
Pd-PPh <sub>2</sub> -MCM-41@SiO <sub>2</sub> @Fe <sub>3</sub> O <sub>4</sub> <sup>a</sup>	0.0683	0.290	530	170	2.8	0.21
Pd-PPh <sub>2</sub> -SiO <sub>2</sub> @Fe <sub>3</sub> O <sub>4</sub> <sup>a</sup>	0.00896	0.195	445	31	1.0	0.030
Pd-PPh <sub>2</sub> -MCM-41 <sup>a</sup>	0.150	0.560	10~1500	510	2.5	0.57
Rh-PPh <sub>2</sub> -MCM-41@SiO <sub>2</sub> @Fe <sub>3</sub> O <sub>4</sub>	0.0765	0.496	530	170	2.8	0.21
Rh-PPh <sub>2</sub> -SiO <sub>2</sub> @Fe <sub>3</sub> O <sub>4</sub>	0.0120	0.108	445	31	1.0	0.030
Rh-PPh <sub>2</sub> -MCM-41	0.167	0.510	10~1500	510	2.5	0.57

<sup>a</sup> The catalysts after being reused for 8 times.

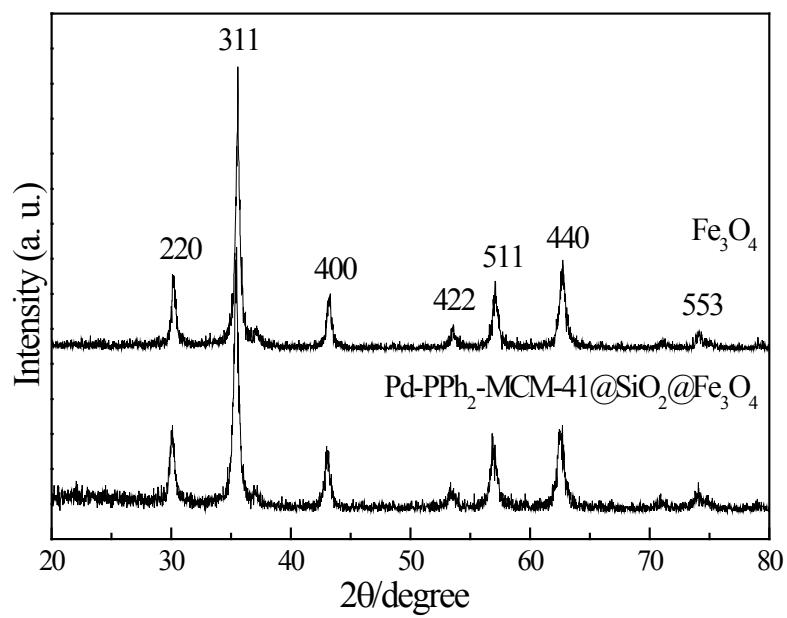


Figure S1 Wide-angle XRD patterns of  $\text{Fe}_3\text{O}_4$  and  $\text{Pd-PPh}_2\text{-MCM-41@SiO}_2@\text{Fe}_3\text{O}_4$ .

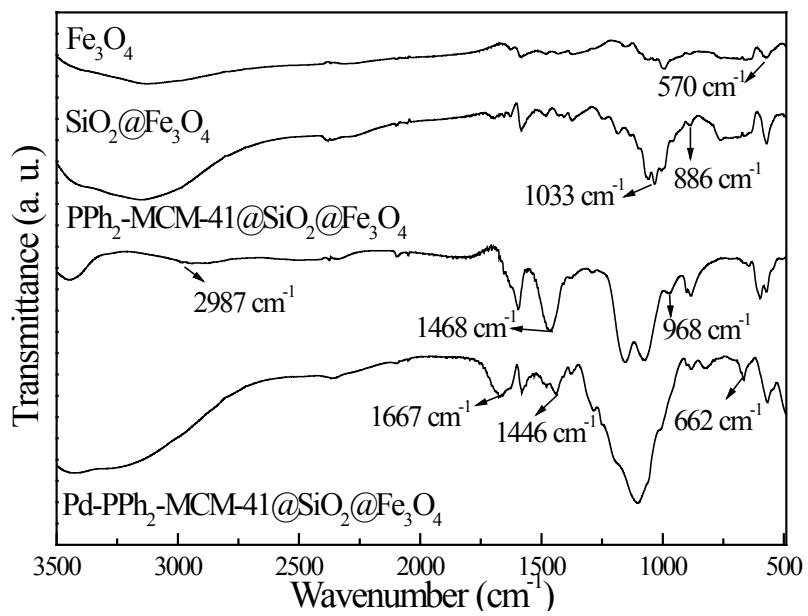


Figure S2 FT-IR spectra of  $\text{Fe}_3\text{O}_4$ ,  $\text{SiO}_2@\text{Fe}_3\text{O}_4$ ,  $\text{PPh}_2\text{-MCM-41}@{\text{SiO}_2@\text{Fe}_3\text{O}_4}$ , and  $\text{Pd-PPh}_2\text{-MCM-41}@{\text{SiO}_2@\text{Fe}_3\text{O}_4}$  samples.

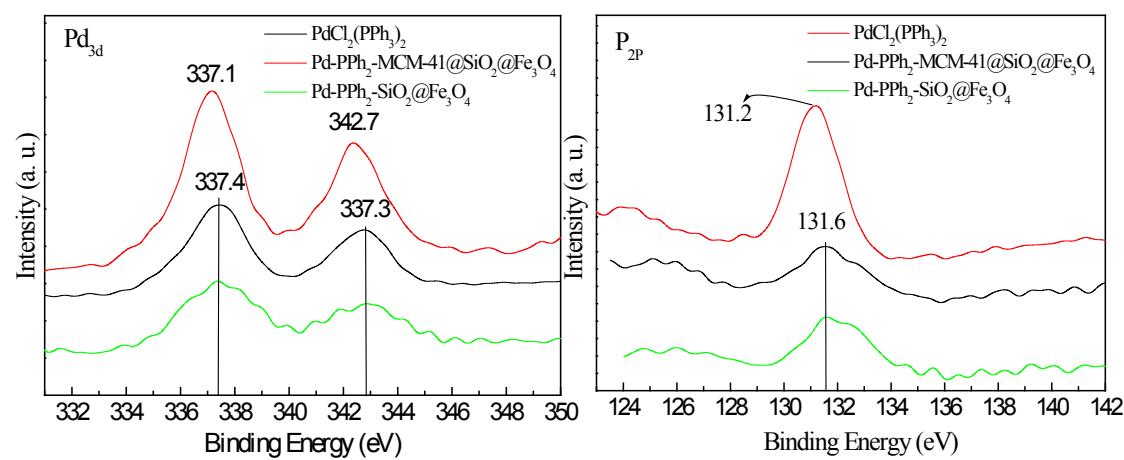


Figure S3 XPS spectra of Pd-PPh<sub>2</sub>-MCM-41@SiO<sub>2</sub>@Fe<sub>3</sub>O<sub>4</sub>, Pd-PPh<sub>2</sub>-SiO<sub>2</sub>@Fe<sub>3</sub>O<sub>4</sub>, and PdCl<sub>2</sub>(PPh<sub>3</sub>)<sub>2</sub>.

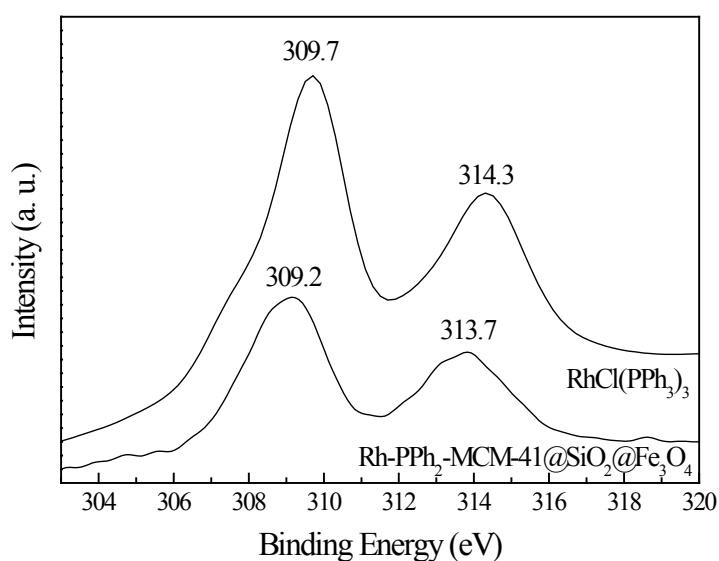


Figure S4 XPS spectra of  $\text{RhCl}(\text{PPh}_3)_3$  and  $\text{Rh-PPh}_2\text{-MCM-41}@{\text{SiO}}_2@{\text{Fe}}_3{\text{O}}_4$  catalysts.

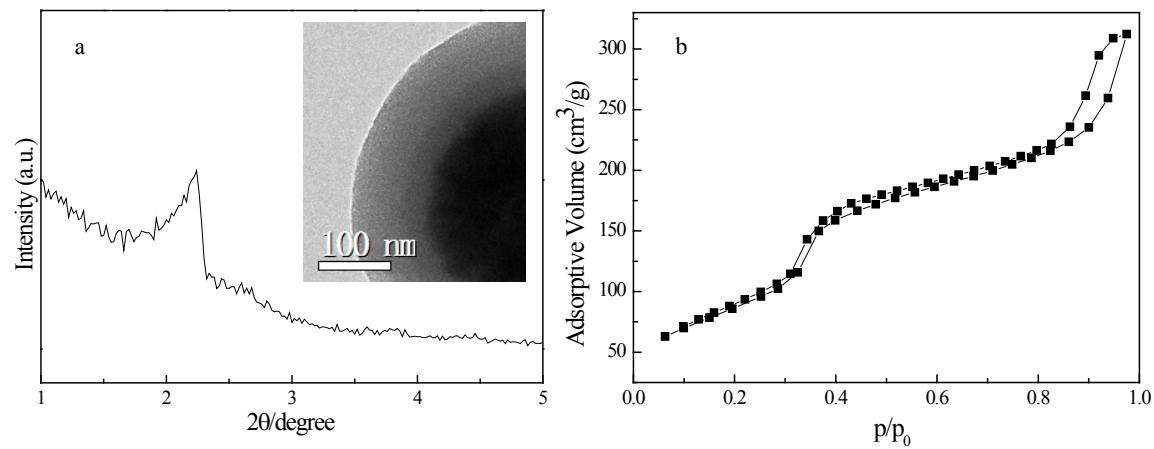


Figure S5 (a) Low-angle XRD pattern and (b)  $\text{N}_2$  sorption isotherm of the Rh-PPh<sub>2</sub>-MCM-41@SiO<sub>2</sub>@Fe<sub>3</sub>O<sub>4</sub>. The attached is the HRTEM image.

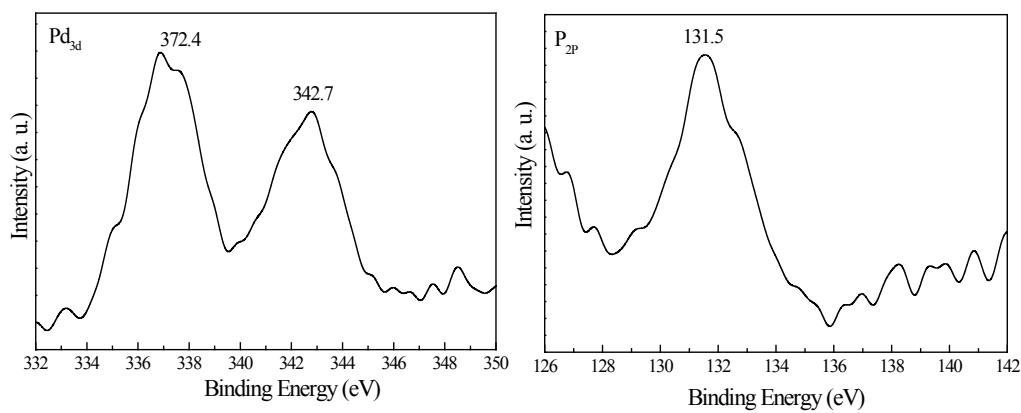


Figure S6 XPS spectra of the Pd-PPh<sub>2</sub>-MCM-41@SiO<sub>2</sub>@Fe<sub>3</sub>O<sub>4</sub> catalyst after being reused repetitively for 8 times.

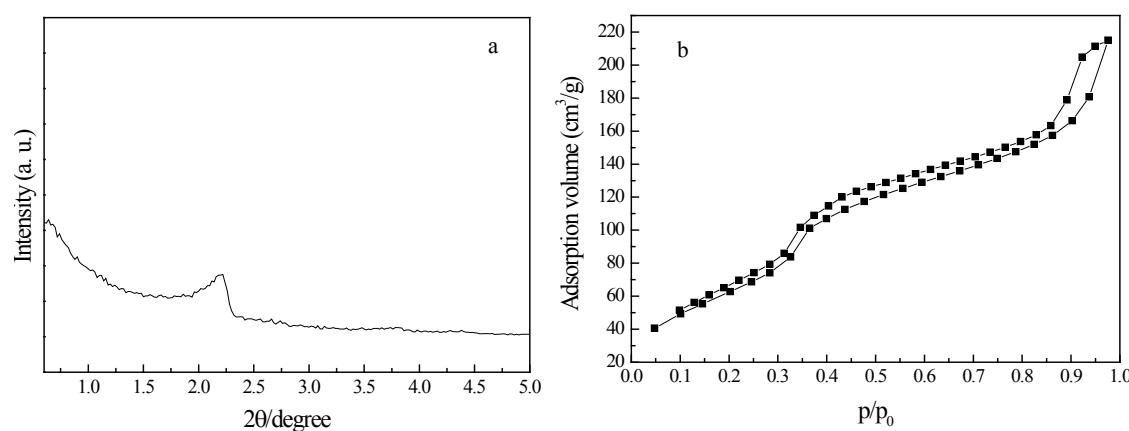


Figure S7 Low-angle XRD pattern (a) and  $\text{N}_2$  adsorption-desorption curve (b) of the Pd-PPh<sub>2</sub>-MCM-41@SiO<sub>2</sub>@Fe<sub>3</sub>O<sub>4</sub> catalyst after being reused repetitively for 8 times.