Supplementary Information

In-situ growth of Co_3O_4 nanoparticles on α -MnO₂ nanotubes: A new hybrid for high-performance supercapacitor

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Fig. S1 Cyclic voltammetric curves (a) and galvanostatic charge-discharge curves (b) of carbon paper; cyclic voltammetric curves (c) and galvanostatic charge-discharge curves (d) of Ni foam. The current density of carbon paper decreased with the repeating scans. In contrary, repeating the cycles led to the increase of current density for Ni foam. Carbon paper hardly contributed to the capacitance, while Ni foam itself was active and gave very big contribution to the capacitance. The calculated capacitance of Ni foam devoted was about 20 $F/(g \cdot cm^2)$, 43 $F/(g \cdot cm^2)$, 92 $F/(g \cdot cm^2)$ and 188 $F/(g \cdot cm^2)$ as the loading amount of active materials was considered as 1 mg, 0.5 mg, 0.75 mg and 0.25 mg, respectively. According to the results, we could demonstrate that it was reasonable to use carbon paper as current collector rather than Ni foam when the loading amount of active materials was 0.8 mg.



Fig S2 SEM images of obtained materials without the addition of NH_4F at high concentration (a-c) (the white arrows in (a-c) show the by-product generated in the synthesis) and low concentration (d) of the precursor solution. TEM images (e, f) of resulting products without the addition of NH_4F at high concentration (black arrows show nanowire clusters and nanofilms).



Fig S3 SEM images of products without the addition of α -MnO₂ nanotubes and NH₄F (a)(b). XRD of resulting materials without the addition of α -MnO₂ nanotubes and NH₄F (curve a) and without the addition of NH₄F (curve b). It indicates that the by-product contains Co(CO₃)_{0.5}(OH)·0.11H₂O in the absence of NH₄F.



Fig. S4 Cyclic voltammetric curves of α -MnO₂ nanotube (a), MnO₂@Co₃O₄-H (b), MnO₂@Co₃O₄-L (c) and MnO₂+Co₃O₄ (d) at different scan rates; galvanostatic discharge curves of α -MnO₂ nanotube (e), MnO₂@Co₃O₄-H (f) and MnO₂+Co₃O₄ (g) at different current densities.