**Electronic Supplementary Information** 

## Nanoparticle-aggregated CuO nanoellipsoids for high performance non-enzymatic glucose detection

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Fig. S1 Size distribution diagram of the width along the longest axis of the CuO nanoellipsoids.



Fig. S2 EDX spectrum of the as-prepared CuO nanoellipsoids.



Fig. S3 XRD patterns of the products obtained at different reaction time: (a) 1 min, (b) 2 min, (c) 5 min, (d) 10 min



Fig. S4 HRTEM image of the CuO nanoparticle obtained at  $V_{NaOH}$  = 3mL (reaction time: 1min).



Fig. S5 HRTEM image of the precursor nanoparticle obtained at  $V_{NaOH} = 3mL$  (reaction time: 1min).



**Fig. S6** XRD patterns of the products obtained at different molar ratio of  $Cu^{2+}$ :  $SnO_3^{2-}$  (a) Sample A, (b) Sample B, (c) Sample C, (d) Sample D.



Fig. S7 XRD patterns of the products obtained at different NaOH: (a) Sample E, (b) Sample F, (c) Sample G.



**Fig. S8** (a) TEM images of the product synthesized in the absence of  $Na_2SnO_3$ . (b) CVs sweep curves for CuO manoplates modified electrode in the presence of glucose concentrations between at 0 mM (black line), 1 mM (redline) and 5 mM (blue-line) in 100 mM KOH at scan rate of 50 mV s<sup>-1</sup>. (c) Amperometric response of the CuO nanoplates modified electrode with successive addition of glucose to the 100 mM KOH solution at regular intervals. The applied potential was +0.55 V (vs. Ag/AgCl (sat'd KCl) reference). (d) Corresponding calibration curve of the CuO glucose sensor, Inset is a linear regression of low glucose concentration (0.1mM ~ 1mM).



Fig. S9 Time-current response of the CuO nanoellipsoids modified glucose sensor.



**Fig. S10** Typical nitrogen gas adsorption-desorption isotherms and pore size distribution curves (insets) of the CuO nanostructure: (a) nanoellipsoids (b) nanoplates.

Sample	CuSO <sub>4</sub> (mmol)	Na <sub>2</sub> SnO <sub>3</sub> (mmol)	NaOH (mmol)	
А	3.6	1.2	7.5	
В	1.2	3.6	7.5	
С	1.2	0.4	7.5	
D	0.4	1.2	7.5	
Е	1.2	1.2	12.5	
F	1.2	1.2	25	
G	1.2	1.2	37.5	

Table S1 Detailed experimental conditions of the products

**Table S2** Comparison of the key performance characteristics of some of existing catalysts for enzyme-free electrooxidation of glucose.

Electrodes	Potential (V)	Sensitivity	Linear range	LOD	Reference
	(•)	$(\mu A m M^{-1} cm^{-2})$	(mM)	(µM)	
CuO nanourchins	+0.5	2682	0.1 ~ 3	1.52	14
CuO nanoflowers	+0.5	2657	0.01 ~ 5	1.71	22
CuO nanourchins	+0.5	1634	up to 5	1.97	26
CuO nanocubes-graphene	+0.59	1360	$2\sim4$	0.7	37
Cu <sub>x</sub> O/Cu	+0.5	1620	up to 4.0	49	38
CuO nanoleaf/MCNTs	+0.35	664.3	0.9	5.7	39
CuO/MWCNTs	+0.7	2190	3	0.8	40
CuO nanoellipsoids	+0.55	2555	0.1 ~ 3	0.072	Current work