

Electronic Supplementary Material (ESI) for

An efficient bifunctional catalyst of Fe/Fe₃C carbon nanofibers for rechargeable Li-O₂ Batteries

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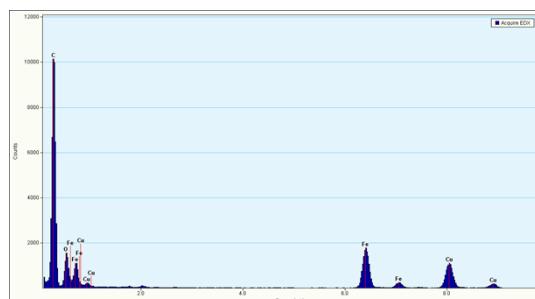


Fig. S1. EDS pattern of Fe/Fe₃C-CNFs.

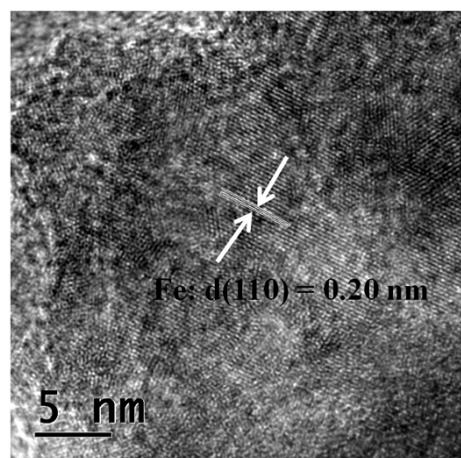


Fig. S2. HR-TEM image of Fe/Fe₃C-CNFs.

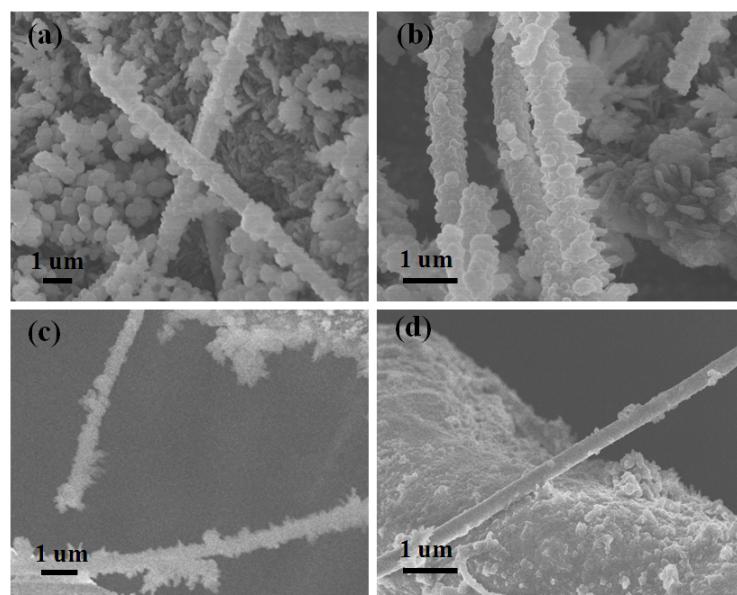


Fig. S3. SEM images of (a)~(c) Fe/Fe₃C-CNF cathode after discharge and (d) Fe/Fe₃C-CNF cathode after recharge.

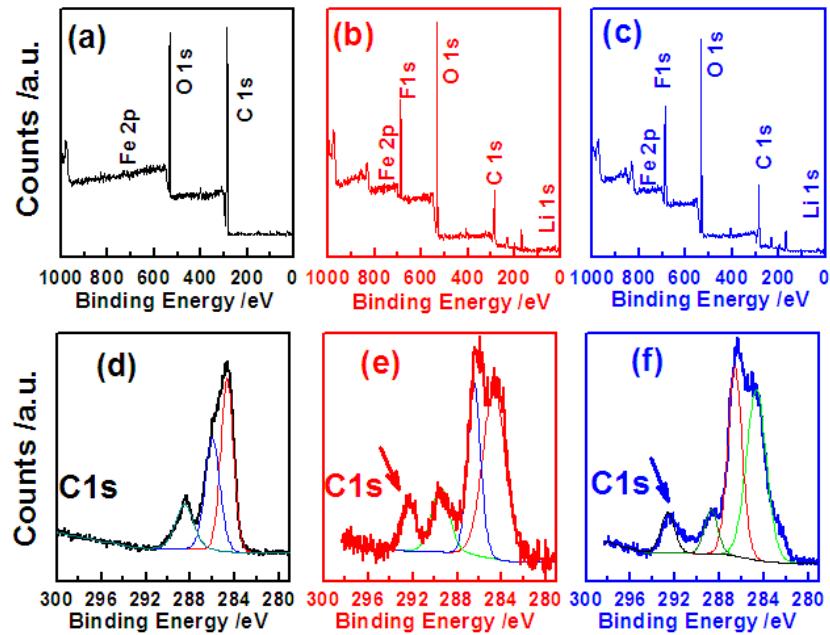


Fig. S4. XPS spectra for Fe/Fe₃C-CNF cathodes marked black for flesh electrode, red for after discharge and blue for recharge: (a)~(c) full spectra; (d)~(f) C 1s.

As revealed in the Li 1s XPS spectra in Fig. 4(d), the main discharge product of Li₂O₂ is generated, and after charge, the Li₂O₂ is decomposed. When switched to the C 1s XPS spectra in Figs. S4(d~f), the signal for C1s at 292.2 eV for Li₂CO₃ can be found in the discharged and recharged Fe/Fe₃C-CNF cathodes.¹⁻² In addition, most of them cannot be decomposed in the charged process.

References

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- 2(a) M. M. Ottakam Thotiyil, S. A. Freunberger, Z. Peng and P. G. Bruce, *Journal of the American Chemical Society*, 2013, **135**, 494; (b) S. A. Freunberger, Y. Chen, N. E. Drewett, L. J. Hardwick, F. Bardé and P. G. Bruce, *Angewandte Chemie International Edition*, 2011, **50**, 8609; (c) Y. Zhang, H. Zhang, J. Li, M. Wang, H. Nie and F. Zhang, *Journal of Power Sources*, 2013, **240**, 390.