

Electronic Supplementary Information (ESI)

A new D-A- π -A type organic sensitizer based on substituted dihydroindolo [2, 3-b] carbazole and DPP unit with a bulky branched alkyl chain for highly efficient DSCs

Guojian Tian,^a Shengyun Cai,^a Xin Li,^b Hans Ågren,^b Qiaochun Wang,^a Jinhai Huang^{a*} and Jianhua Su^{a*}

^a Key Laboratory for Advanced Materials and Institute of Fine Chemicals, East China University of Science & Technology, 130 Meilong Road, Shanghai, 200237, PR China. Fax: 862164252288; E-mail: bbsjh@ecust.edu.cn; huangjh@ecust.edu.cn.

^b Division of Theoretical Chemistry and Biology, School of Biotechnology, KTH Royal Institute of Technology, SE-10691 Stockholm, Sweden.

1. The frontier orbital plots of the HOMO and LUMO of **T1-T2**(**Table 3**);
2. Computational analysis(**Table 4**), (**Fig. 4**);
3. Absorption curves of dyes **T1-T2** upon light irradiation of AM 1.5 solar light (30 min) with UV cutoff filter at 400 nm.
4. Normalized absorption and emission spectra of **T1-T2** in both solvent systems.
5. The absorption amounts of **T1-T2** in CHCl₃: EtOH (v/v 3:7) and CH₂Cl₂ were showed in **Table 5**.
6. The printout of elemental analysis of **T1-T2**.

1. The frontier orbital plots of the HOMO and LUMO of **T1-T2**.

	T1	T2
HOMO-4		
HOMO-2		
HOMO-1		
HOMO		
LUMO		
LUMO+1		
LUMO+2		

Table 3. Contour plots of frontier molecular orbitals of compounds **T1** and **T2**.

2. Computational analysis

Compound	State	λ_{abs}	f	MO composition
T1	S ₁	2.46 eV, 503 nm	1.631	H-2 -> L+0 (15%) H-1 -> L+0 (59%) H-1 -> L+1 (15%)
	S ₂	3.19 eV, 388 nm	0.450	H-4 -> L+0 (46%) H-1 -> L+1 (30%)
	S ₅	3.54 eV, 349 nm	1.145	H-2 -> L+0 (30%) H-2 -> L+1 (16%) H-1 -> L+2 (11%)
T2	S ₁	2.56 eV, 485 nm	1.366	H-2 -> L+0 (15%) H-1 -> L+0 (58%) H-1 -> L+1 (18%)
	S ₅	3.60 eV, 344 nm	1.810	H-2 -> L+0 (18%) H-2 -> L+1 (25%) H-1 -> L+0 (10%)

Table 4. Computed excitation energies, oscillator strengths and molecular orbital (MO) compositions of important low-lying excited states of **T1** and **T2**.

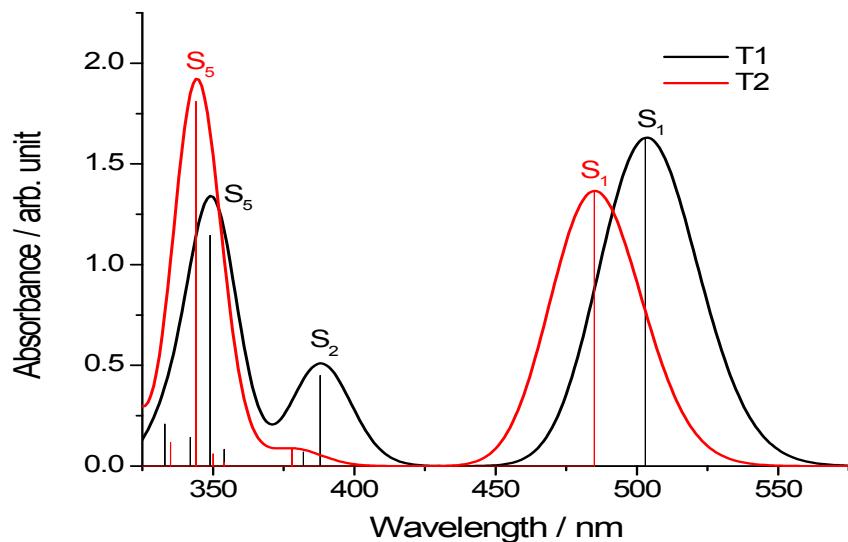
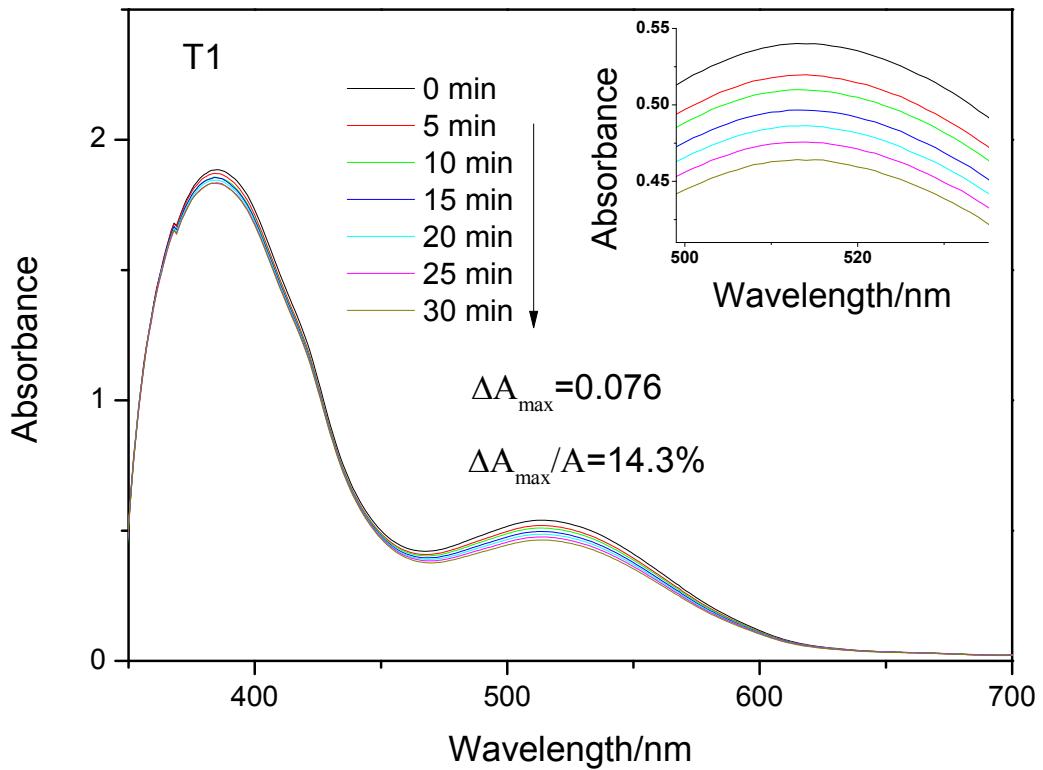
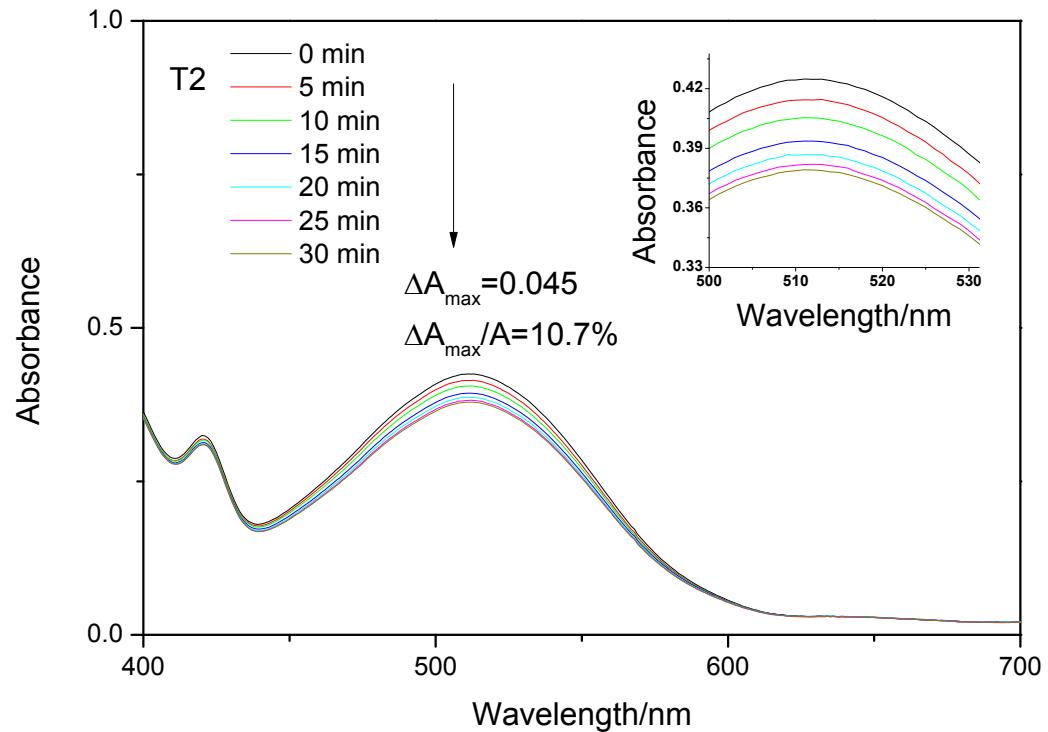


Fig. 4 Simulated absorption spectra of **T1** and **T2**.

3. Absorption curves of dyes **T1-T2** upon light irradiation of AM 1.5 solar light (30 min) with UV cutoff filter at 400 nm.

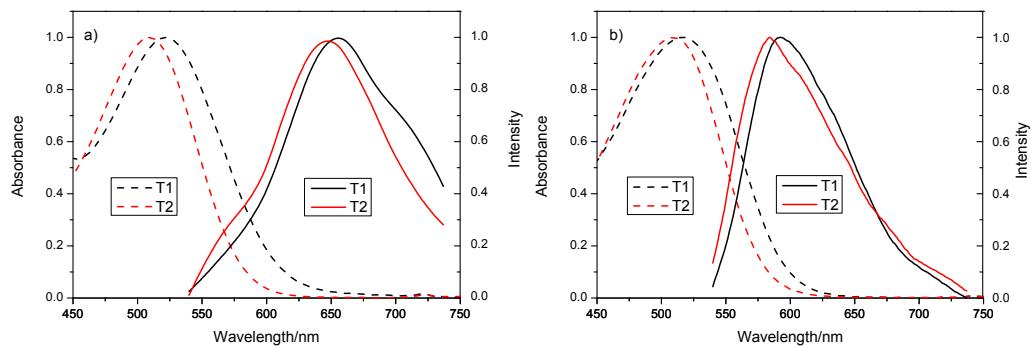


Absorption curves of dyes **T1** upon light irradiation of AM 1.5 solar light (30 min) with UV cutoff filter at 400 nm.



Absorption curves of dyes **T2** upon light irradiation of AM 1.5 solar light (30 min) with UV cutoff filter at 400 nm.

4. Normalized absorption and emission spectra of **T1-T2** in both solvent systems.

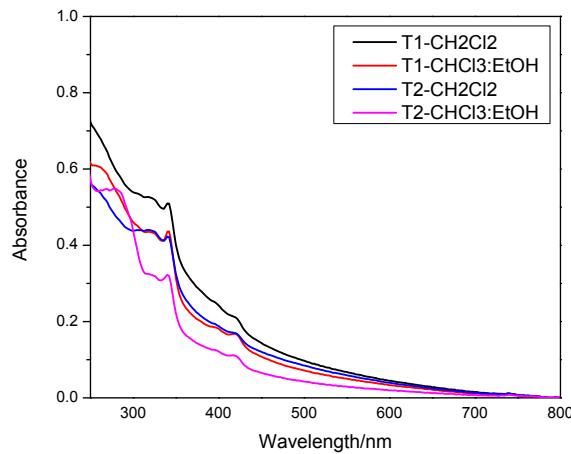


Normalized absorption and emission spectra of **T1-T2** in both solvent systems: a) was in CH_2Cl_2 ; b) was in CHCl_3 : EtOH (v/v 3:7).

5. The absorption amounts of **T1-T2** in CHCl_3 : EtOH (v/v 3:7) and CH_2Cl_2 were showed in Table 5.

T1- CHCl_3 : EtOH	T2- CHCl_3 : EtOH	T1- CH_2Cl_2	T2- CH_2Cl_2
$2.32 \times 10^{-8}(\text{molcm}^{-2})$	$1.94 \times 10^{-8}(\text{molcm}^{-2})$	$2.68 \times 10^{-8}(\text{molcm}^{-2})$	$2.34 \times 10^{-8}(\text{molcm}^{-2})$

Table 5. The absorption amounts of **T1-T2** in CHCl_3 : EtOH (v/v 3:7) and CH_2Cl_2 . The corresponding absorption spectra were below:



6. The printout of elemental analysis of T1-T2.

The analysis and test center in East China university of science and technology																													
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