

Electronic Supplementary Information for

Al-doping to Synchronously Improve Conduction Band and Electron Lifetime for SnO₂ Photoanode to Enhance Dye-Sensitized Solar Cells Performances

Yandong Duan,^{†ab} Jiaxin Zheng,^{†a} Nianqing Fu,^{bc} Yanyan Fang,^b Tongchao Liu,^a Qian Zhang,^d Xiaowen Zhou,^b Yuan Lin,^{*ab} and Feng Pan^{*a}

^a School of Advanced Materials, Peking University, Peking University Shenzhen Graduate School, Shenzhen 518055, China. E-mail: linyuan@iccas.ac.cn; panfeng@pkusz.edu.cn; Tel: 86-0755-26033200.

^b Beijing National Laboratory for Molecular Sciences, Key Laboratory of Photochemistry, Institute of Chemistry, Chinese Academy of Sciences, Beijing 100190, China.

^c Department of Applied Physics, The Hong Kong Polytechnic University, Hung Hom, Kowloon, Hong Kong, China.

^d State Key Laboratory for Advanced Metals and Materials, School of Materials Science and Engineering, University of Science and Technology Beijing, Beijing 100083, China.

† Y. D. Duan and J. X. Zheng contributed equally to this work.

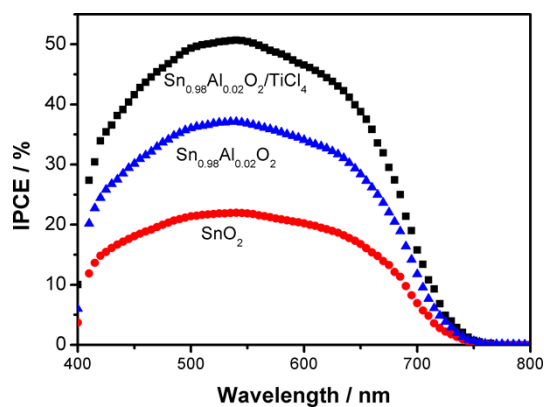


Fig. S1 IPCE of the fabricated DSSCs with SnO_2 , $\text{Sn}_{0.98}\text{Al}_{0.02}\text{O}_2$, and $\text{Sn}_{0.98}\text{Al}_{0.02}\text{O}_2/\text{TiCl}_4$ as photoanodes.

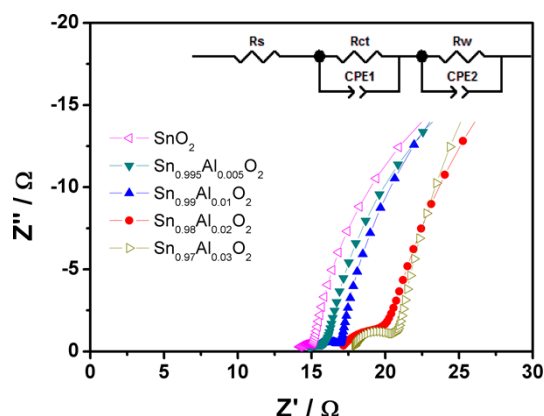


Fig. S2 EIS spectra of the SnO_2 and the Al-doped SnO_2 DSSCs. The inset of EIS plots represents the equivalent circuit for EIS.

Table S1 Comparison for the photovoltaic performance of the DSSCs based on different SnO_2 photoanode structures (SnO_2 , $\text{SnO}_2/\text{TiO}_2$, and SnO_2/ZnO composites).

Ref.	Morphology or structure	Diameter	DSSCs based on SnO_2 photoanode			
			Synthetic method or manufacturer	Film thickness	η (%) (No surface treatment)	η (%) (After surface treatment) ^a
S1	SnO_2 nanoparticles	10-30 nm	NanoTek, SNW15WT%-G02	6 μm	2.3	---
S2	SnO_2 nanoparticles	3-5 nm	Alfa Aesar	10 μm	1.74	MgO/7.21
S3	SnO_2 nanowire	20-200 nm	reactive vapor transport	25-30 μm	2.1	TiCl_4 /4.1
S4	SnO_2 hollow microspheres	1-2 μm	hydrothermal	10 μm	1.4	TiCl_4 /5.65
S5	coral-like SnO_2	300 nm \times 2 μm	wet-chemical	---	1.04	---
S6	SnO_2 nanoparticles	15 nm	Alfa Aesar	4 μm	0.76	Al_2O_3 /3.7
S7	meso- SnO_2	20 nm pores	hard template method	3 μm	1.1	TiCl_4 /3.8
S8	SnO_2 nanoflower	1 μm	hydrothermal	8-10 μm	3.00	TiCl_4 /6.78

S9	SnO ₂ nanoparticles	15 nm	Alfa Aesar	5 μm	1.14	NiO/1.85
S10	SnO ₂ nanoparticles	15 nm	Alfa Aesar	---	1.7	CaCO ₃ /5.4
S11	nanoporous SnO ₂	4 nm/>100 nm	Alfa Aesar /Aldrich	1-5μm	2.27	---
S12	SnO ₂ nanocrystals	100 nm	microwave solvothermal	10-12μm	3.16	---
S13	SnO ₂ nanoarborous structure	---	electrodeposition	15μm	0.47	---
S14	SnO ₂ nanopowder	<100 nm	Sigma-Aldrich	8μm	3.65	MgO/6.40
S15	SnO ₂ nanotube	110 nm	electrospinning	13μm	0.99	TiCl ₄ /5.11
S16	SnO ₂ nanowires	75±25 nm	electrospinning	19±2 μm	2.53	---
S17	Zn-doped SnO ₂ nano-echinus	1 μm	solvothermal	11μm	4.15	---
S18	SnO ₂ nanoparticles	100 nm	hydrothermal	---	0.85	4.15
S19	SnO ₂ hollow nanospheres	200 nm	hydrothermal	---	0.86	TiCl ₄ /6.02
S20	SnO ₂ octahedra	0.5-1.8 μm	sonochemical	13.2 μm	---	TiCl ₄ /6.8
S21	SnO ₂ nanosheet	thickness:4-6 nm	hydrothermal	4.1μm	0.23	TiCl ₄ /1.79
S22	mesoporous SnO ₂ agglomerates	200-600 nm	molten salt method	8 μm	3.05	TiCl ₄ /6.23
S23	N-SnO ₂ mesoporous microspheres	1.2-1.5 μm	one-pot solvothermal	---	2.3	---
S24	Sb-doped SnO ₂ aerogels cauliflower-like	---	sol-gel	10μm	0.7	ALD TiO ₂ /3.5
S25	SnO ₂ hollow microspheres	500 nm-2 μm	hydrothermal	11μm	---	3.6
S26	SnO ₂ nanofibers	200 nm	---	8.7 μm	--	TiCl ₄ /4.63
S27	SnO ₂ nanoparticles	11.2-26.2 nm	microwave hydrothermal	13-15 μm	1.35	---
S28	SnO ₂ nanoflower	1μm	hydrothermal	--	1.05	TiCl ₄ /5.6
Our work	Al-SnO₂ nanocrystals	11.6-15.9 nm	hydrothermal	8 μm	3.56	TiCl₄/6.91

DSSCs based on SnO₂/TiO₂ and SnO₂/ZnO composites

S26	SnO ₂ /TiO ₂ composite (1:1)	---	mechanical blend	7.5 μm	---	TiCl ₄ /6.17
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S29	SnO ₂ nanorod@TiO ₂	150 nm×40 nm	flame spray pyrolysis	12 μm	3.95	TiCl ₄ /6.98
S30	SnO ₂ nanoparticles/ZnO nanotetrapods	6-10 nm/40×500 nm	hydrothermal/metal vapor transport- oxidation method	6 μm	6.31	---
S31	SnO ₂ nanoparticle-ZnO nanorod	---/103-291 nm×7μm	hydrothermal	3 μm +4.4 μm	2.62	---
S32	SnO ₂ NRs-TiO ₂	50 nm×5 nm	solution method	150nm+1 0μm	8.61	---
S33	SnO ₂ hollow spheres-TiO ₂ nanosheets	500 nm	solvothetical reaction	8μm	8.2	---

^aSurface treatment method and the corresponding photon-to-electron conversion efficiency.

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