Supporting Information

Synthesis, optical properties and charge transport characteristics of a series of novel thiophene-fused phenazine derivatives

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Figure S1. Cyclic voltammograms (scan rate 100 mV s⁻¹) of 100nm thin film of *l*-**TTP** and *m*-**TTP** on ITO recorded in DCM solution containing (n-Bu₄N)PF₆ (0.1M) as electrolyte. ITO and a Pt electrode were used for the working electrode and the counter electrode, respectively. Potentials referenced to Ag/Ag⁺.



Figure S2. Out-of-plane XRD pattern of *l*-HTTP, *t*-HTTP and *m*-HTTP vacuum-sublimed on Si substrates (no peak in in-plane spectrum) with $T_{sub} = 75^{\circ}$ C.



Figure S3. Out-of-plane XRD pattern (Cu $K\alpha$ source: $\lambda = 1.541$ Å) of *l*-**TTP**, *m*-**TTP** and *t*-**TTP** vacuum-sublimed on Si substrates (no peak in in-plane spectrum) at $T_{sub} = 25$ °C.

	Observed absorption band (λ_{max})	Calculated by TD-DFT	Assigned transition ^a	Energy difference	Contri- bution	Oscillator strength (f)
	302 nm	286 nm	$HOMO \rightarrow LUMO + 2$	4.56 eV	0.24	0.831
<i>l</i> -TTP			HOMO - 2→LUMO	3.74 eV	0.64	
	408 nm	375 nm	HOMO→LUMO + 2 HOMO - 2→LUMO	4.56 eV 3.74 eV	0.62 -0.16	0.399
	212	200	$HOMO \rightarrow LUMO + 1$	4.42 eV	0.58	0.201
TTD	512 nm	309 nm	HOMO - 1→LUMO	3.50 eV	-0.12	0.381
<i>m</i> -11r			$HOMO \rightarrow LUMO + 1$	4.42 eV	0.22	
	439 nm	405 nm	HOMO - 1→LUMO	3.50 eV	0.64	0.325
	220	211	HOMO - $1 \rightarrow LUMO + 1$	4.37 eV	0.63	0 707
t-TTP	320 nm 311	311 nm	HOMO→LUMO	3.29 eV	0.13	0.797
, , , ,		HOMO - $1 \rightarrow$ LUMO + 1	4.37 eV	-0.22	0.000	
	467 nm	432 nm	HOMO→LUMO	3.29 eV	0.64	0.302
	300 nm	206	$HOMO \rightarrow LUMO + 2$	4.63 eV	0.59	1.446
<i>l</i> -TTA	(strong)	286 nm	HOMO - 1→LUMO	4.13 eV	0.35	1.446
	479 nm	250	$HOMO \rightarrow LUMO + 2$	4.63 eV	-0.35	0.105
	(weak)	350 nm	HOMO - 1→LUMO	4.13 eV	0.61	0.127
	310 nm	220	$HOMO \rightarrow LUMO + 1$	4.09 eV	0.60	1.2.62
t-TTA	(strong)	320 nm	HOMO - 1→LUMO	3.77 eV	-0.37	1.262
	418 nm	202	$HOMO \rightarrow LUMO + 1$	4.09 eV	0.37	0.059
	(weak)	393 nm	HOMO - 1→LUMO	3.77 eV	0.60	0.058

Table S1. Absorption wavelengths and their theoretical assignments for *l*-TTP, *m*-TTP, *t*-TTP, *l*-TTA, and *t*-TTA.

^{*a*} Assigned transitions and their corresponding energy difference were showed in Figure 4.



Figure S4. Transfer (left) and output (right) characteristics of thin films of *l*-**HTTP** (a), *m*-**HTTP** (c) and *t*-**HTTP** (d) vacuum-sublimed on SiO₂/Si. Thin film of *l*-**HTTP** on HMDS treated SiO₂/Si (b).



Figure S5. Conductivity change during I_2 vapor doping in thin film (a) and crystal (b) of *l*-**TTP**; and thin film (c) and crystal (d) of *l*-**HTTP** (conductivity data of the other derivatives are listed in Table 4).



Figure S6. (a) Energy diagram and distributions of frontier orbitals of *l*-**TTP** accompanied with the assignment of optical transitions. (b) UV-vis spectrum of *l*-**TTP** (red line) in CHCl₃ accompanied with a simulated transitions (black line) calculated by TD-DFT method (B3LYP/6-31G (d)). Scalling factor is 0.94.



Figure S7. (a) Energy diagram and distributions of frontier orbitals of *m*-**TTP** accompanied with the assignment of optical transitions. (b) UV-vis spectrum of *m*-**TTP** (red line) in CHCl₃ accompanied with a simulated transitions (black line) calculated by TD-DFT method (B3LYP/6-31G (d)). Scalling factor is 0.94.



Figure S8. *t***-TTP**: (a) Energy diagram and distributions of frontier orbitals of *t***-TTP** accompanied with the assignment of optical transitions. (b) UV-vis spectrum of *t***-TTP** (red line) in CHCl₃ accompanied with a simulated transitions (black line) calculated by TD-DFT method (B3LYP/6-31G (d)). Scalling factor is 0.94.



Figure S9. (a) Energy diagram and distributions of frontier orbitals of *l*-HTTP accompanied with the assignment of optical transitions. (b) UV-vis spectrum of *l*-HTTP (red line) in CHCl₃ accompanied with a simulated transitions (black line) calculated by TD-DFT method (B3LYP/6-31G (d)). Scalling factor is 0.94.

0.900

0.378



Observed absorption band (λ_{max})	Calculated by TD-DFT	Assigned transition	Energy difference	Contribution	Oscillator strength (f)
215	319 nm	HOMO→LUMO + 1	4.30 eV	0.62	0.(27
315 nm		HOMO - 1→LUMO	3.58 eV	0.13	0.627
451 mm	414 mm	$HOMO \rightarrow LUMO + 1$	4.30 eV	-0.21	0 291
431 1111	414 nm	HOMO - 1→LUMO	3.58 eV	0.64	0.381

Figure S10. (a) Energy diagram and distributions of frontier orbitals of *m*-HTTP accompanied with the assignment of optical transitions. (b) UV-vis spectrum of *m*-HTTP (red line) in CHCl₃ accompanied with a simulated transitions (black line) calculated by TD-DFT method (B3LYP/6-31G (d)). Scalling factor is 0.94.



band (λ_{max})	by TD-DFT	Assigned transition	difference	Contribution	strength (<i>f</i>)
328 nm	321 nm	HOMO - $1 \rightarrow$ LUMO + 1	4.47 eV	0.64	1 178
	521 mm	HOMO→LUMO	3.21 eV	0.13	1.170
486 nm	445 nm	HOMO - $1 \rightarrow$ LUMO + 1	4.47 eV	-0.21	0.424
		HOMO→LUMO	3.21 eV	0.64	0.434

Figure S11. (a) Energy diagram and distributions of frontier orbitals of *t*-HTTP accompanied with the assignment of optical transitions. (b) UV-vis spectrum of *t*-HTTP (red line) in CHCl₃ accompanied with a simulated transitions (black line) calculated by TD-DFT method (B3LYP/6-31G (d)). Scalling factor is 0.94.



Observed absorption band (λ_{max})	Calculated by TD-DFT	Assigned transition	Energy difference	Contribution	Oscillator strength (<i>f</i>)
200 (()	296	$HOMO \rightarrow LUMO + 2$	4.63 eV	0.59	1 446
300 nm(strong)	286 nm	HOMO - 1→LUMO	4.13 eV	0.35	1.446
470 mm(weak)	250 mm	$HOMO \rightarrow LUMO + 2$	4.63 eV	-0.35	0 127
479 mm(weak)	550 mm	HOMO - 1→LUMO	4.13 eV	0.61	0.127

Figure S12. *l***-TTA**: (a) Energy diagram and distributions of frontier orbitals of *l***-TTA** accompanied with the assignment of optical transitions. (b) UV-vis spectrum of *l***-TTA** (blue line) in DCM¹ accompanied with a simulated transitions (black line) calculated by TD-DFT method (B3LYP/6-31G (d)). Scalling factor is 0.94.



Observed absorption band (λ_{max})	Calculated by TD-DFT	Assigned transition	Energy difference	Contribution	Oscillator strength (f)
210	ong) 320 nm	$HOMO \rightarrow LUMO + 1$	4.09 eV	0.60	1.262
310 nm (strong)		HOMO - 1→LUMO	3.77 eV	-0.37	1.262
418 nm (weak)	393 nm	$HOMO \rightarrow LUMO + 1$	4.09 eV	0.37	0.059
		HOMO - 1→LUMO	3.77 eV	0.60	0.058

Figure S13. *t***-TTA**: (a) Energy diagram and distributions of frontier orbitals of *t***-TTA** accompanied with the assignment of optical transitions. (b) UV-vis spectrum of *t***-TTA** (purple line) in DCM¹ accompanied with a simulated transitions (black line) calculated by TD-DFT method (B3LYP/6-31G (d)). Scalling factor is 0.94.



band (λ_{max})	by TD-DFT	Assigned transition	difference	Contribution	strength (f)
480 nm	410 mm	LUMO→HOMO - 2	3.56 eV	0.69	0.642
	410 nm	LUMO + 3→HOMO		-0.15	0.642

Figure S14. (a) Energy diagram and spatial distributions of frontier orbitals of an excited state of *l*-**TTP** accompanied with the assignment of photoluminescence transition. (b) Observed fluorescence spectrum (red line) of in CHCl₃ solution ($\lambda_{ex} = 303$ nm) and simulated spectrum (black line) with TD-DFT method (B3LYP/6-31G (d)). Scalling factor is 0.94.



Observed absorption band (λ_{max})	Calculated by TD-DFT	Assigned transition	Energy difference	Contribution	Oscillator strength (f)
491	119	LUMO→HOMO - 1	3.50 eV	0.69	0.502
481 nm	448 nm	LUMO + 1→HOMO	4.22 eV	-0.12	0.503

Figure S15. (a) Energy diagram and spatial distributions of frontier orbitals of an excited state of *m*-**TTP** accompanied with the assignment of photoluminescence transition. (b) Observed fluorescence spectrum (red line) of in CHCl₃ solution ($\lambda_{ex} = 316$ nm) and simulated spectrum (black line) with TD-DFT method (B3LYP/6-31G (d)). Scalling factor is 0.94.



Figure S16. (a) Energy diagram and spatial distributions of frontier orbitals of an excited state of *t*-**TTP** accompanied with the assignment of photoluminescence transition. (b) Observed fluorescence spectrum (red line) of in CHCl₃ solution ($\lambda_{ex} = 322$ nm) and simulated spectrum (black line) with TD-DFT method (B3LYP/6-31G (d)). Scalling factor is 0.94.



Figure S17. Effective intermolecular contacts in *l***-TTP** crystal along *b*-axis (a), *b*-axis (b) and a + c direction (c).

Table S2. Theoretical estimated intermolecular transfer intergrals (*Hab*), overlap intergrals (*Sab*), center-to-center distances (*d*), orbital interaction energy (*V*), reorganization energy (λ) and hopping mobilities for each molecular contact.

	Contact ^a	Hab^{b} /meV	Sab^b /meV	d /Å	V/meV	λ^{c} / meV	$\frac{\mu_{\text{hopping}}}{\text{cm}^2 \text{ V}^{-1} \text{ S}^{-1}}$
<i>l</i> -TTP	b	-135.4	11.2	3.892	-76.5	153	5.3×10^{-1}
<i>l</i> -TTP	а	7.7	-0.7	8.748	.3.6	153	6.1×10^{-3}
<i>l</i> -TTP	С	-8.9	-0.9	13.001	4.0	153	1.7×10^{-2}
<i>l</i> -TTP	с'	14.47	-1.6	13.002	.5.9	153	$3.5 imes 10^{-2}$
<i>l</i> -TTP	a+c	-1.5	0.2	12.934	0.6	153	$4.0 imes 10^{-4}$
<i>l</i> -TTP	(a+c)'	3.2	-0.3	13.507	.1.6	153	$2.9 imes 10^{-3}$
<i>l</i> -TTP	b'	-162.4	13.9	3.892	91.3	153	$7.6 imes 10^{-1}$

^{*a*} Corresponding molecular contacts are indicated in figure S17. ^{*b*} Calculated in PW91/TZ2P level.

^{*c*} Calculated in B3LYP/6-31G(d) level.



Figure S18. Effective intermolecular contacts in *m*-**TTP** crystal along *c*-axis (a), *a*-axis (b) and a + c direction (c).

Table S3. Theoretical estimated intermolecular transfer intergrals (*Hab*), overlap intergrals (*Sab*), center-to-center distances (*d*), orbital interaction energy (*V*), reorganization energy (λ) and hopping mobilities for each molecular contact.

	Contact ^a	Hab^b /meV	Sab^b /meV	d /Å	V/meV	λ^{c} / meV	$\frac{\mu_{\rm hopping}}{\rm cm^2 V^{-1} S^{-1}}$
<i>m</i> -TTP	а	-0.7	0.1	12.581	-0.3	169	$8.9 imes 10^{-5}$
<i>m</i> -TTP	a+b	0.4	0	13.474	0.2	169	$4.0 imes 10^{-5}$
<i>m</i> -TTP	b	-133.3	12.9	4.825	-66.5	169	$5.0 imes10^{-1}$
<i>m</i> -TTP	C//	-1.0	0	7.857	-1.1	169	$3.7 imes 10^{-4}$
<i>m</i> -TTP	c//'	-1.8	0	7.857	-1.9	169	$1.1 imes 10^{-3}$
<i>m</i> -TTP	\mathcal{C}_{\perp}	25.9	-2.6	11.941	11.7	169	$9.5 imes 10^{-2}$
<i>m</i> -TTP	c⊥'	25.7	-2.7	11.941	11.5	169	9.2×10^{-2}

^{*a*} Corresponding molecular contacts are indicated in figure S18. ^{*b*} Calculated in PW91/TZ2P level.

^c Calculated in B3LYP/6-31G(d) level.



Figure S19. Effective intermolecular contacts in *t***-TTP** crystal along *c*-axis (a), *b*-axis (b) and a - c direction (c).

Table S4. Theoretical estimated intermolecular transfer intergrals (*Hab*), overlap intergrals (*Sab*), center-to-center distances (*d*), orbital interaction energy (*V*), reorganization energy (λ) and hopping mobilities for each molecular contact.

	Contact ^a	Hab^b /meV	Sab^b /meV	d /Å	V/meV	λ^{c} /m eV	$\frac{\mu_{\text{hopping}}}{\text{cm}^2 \text{ V}^{-1} \text{ S}^{-1}}$
<i>t</i> -TTP	2 <i>b</i> - <i>a</i>	0.2	-0.02	15.069	0	174	1.9×10^{-6}
t-TTP	b-a	0.6	-0.2	12.314	-0.4	174	$1.3 imes 10^{-4}$
t-TTP	b	59.1	-7.4	5.014	20.2	174	4.7×10^{-2}
t-TTP	a+c	-21.9	2.4	9.262	-9.3	174	3.4×10^{-2}
t-TTP	(a+c)'	-24.1	2.4	9.262	-9.3	174	$5.0 imes 10^{-2}$
t-TTP	а-с	-8.3	0.9	11.862	-3.6	174	8.4×10^{-3}
t-TTP	(a-c)'	5.0	-0.4	9.509	3.0	174	3.7×10^{-3}

^{*a*} Corresponding molecular contacts are indicated in figure S19. ^{*b*} Calculated in PW91/TZ2P level. ^{*c*} Calculated in B3LYP/6-31G(d) level.



Figure S20. Effective intermolecular contacts in *l*-HTTP crystal.

Table S5. Theoretical estimated intermolecular transfer intergrals (*Hab*), overlap intergrals (*Sab*), center-to-center distances (*d*), orbital interaction energy (*V*), reorganization energy (λ) and hopping mobilities for each molecular contact.

	Contact ^a	Hab ^b /meV	Sab^b /meV	d /Å	V/meV	$\lambda^{c}/m eV$	$\frac{\mu_{\text{hopping}}}{\text{cm}^2 \text{ V}^{-1} \text{ S}^{-1}}$
<i>l</i> -HTTP	t_1	-50.8	5.6	4.892	-25.0	155	$8.7 imes 10^{-2}$
<i>l</i> -HTTP	t_3	-0.4	0	14.988	0	155	$6.3 imes 10^{-5}$
<i>l</i> -HTTP	t_2	-3.9	0.4	12.749	-2.0	155	$3.8 imes 10^{-3}$

^{*a*} Corresponding molecular contacts are indicated in figure S20. ^{*b*} Calculated in PW91/TZ2P level. ^{*c*} Calculated in B3LYP/6-31G(d) level.

¹ Brusso, J. L.; Hirst, O. D.; Dadvand, A.; Ganesan, S.; Cicoira, F.; Robertson, C. M.; Oakley, R. T.; Rosei, F.; Perepichka, D. F. *Chem. Mater.* **2008**, *20*, 2484-2494.