

Supporting information

A facile synthesis of phase-pure FeCr_2Se_4 and FeCr_2S_4 nanocrystals via wet chemistry method

Xiang Mao, Jaebeom Lee*

Department of Nano Fusion Engineering, and Cogno-Mechatronics
Engineering, Pusan National University, Busan 609-735, Republic of
Korea

Corresponding author: jaebeam@pusan.ac.kr

Conductivity mechanism:

A simplified expression of conductivity can be obtained by converting the previous equation. As shown in Debye's isotropic continuum model,¹⁻³ the resulting expression is:

$$I = H^2 Q_d^5 / 6\pi^2 K_b T^2 \sum_s^4 (C_s^4) \int_0^1 dx x^4 \Gamma \bar{n} (\bar{n} + 1) \quad (1)$$

In equation (1), where $x = Q/Q_d$ and Γ and \bar{n} understood as functions of x , T and

polarization s . The total relaxation is $\Gamma_{qs}^{-1} = \sum \Gamma_{qs}^{-1}(i)$, with $\Gamma_{qs}^{-1}(i)$ being the

relaxation due to the phonon mechanism. The conductivity can be approximate as:

$$I = (H^2 / 3N_0 \Omega K_B T^2) \sum_{qs}^0 \sigma C_s^2(q) \omega^2(qs) \tau(qs) n_{qs} (n_{qs} + 1) \quad (2)$$

$$\sigma = \lambda_L / L * T \quad (3-1)$$

$$\Delta \Omega = \Delta W * V * L_s^{1/2} / W * V = \Delta W * L_s^{1/2} / W \quad (3-2)$$

In equation (2), provided reasonable forms of the phonon-dispersion relation $\omega = \omega(qs)$ and relaxation time $\tau(qs)$ are chosen, it should be straightforward to be proved by the conductivity of a crystal lattice contribution (λ_L). The nature of conductivity is dependent on the NCs's λ_L , where L is the Lorenz factor, σ stands for the extent of NCs' crystallization; T is the temperature value in conductivity measurements (3-1).

The related resistivity $\Delta \Omega$ which is dependent on the NCs' size parameter (L_s), where ΔW is the increase over the crystal resistivity W , L_s is the average grain size, V is the volume of NCs (3-2).

References:

- [1] Sun, Y.; Rogers, J. A. Inorganic semiconductors for flexible electronic. *Adv Mater.* 2007, **19**, 1897.
- [2] Watari, K.; Shinde, S. L. High thermal conductivity materials. *MRS Bulletin.* 2001, **26**, 440.

[3] Watari, K.; Hirao, K.; Toriyama, M.; Ishizaki, K. Effect of grain size on the thermal conductivity of Si_3N_4 . *J. Am. Ceram. Soc.* 1999, **82**, 777.

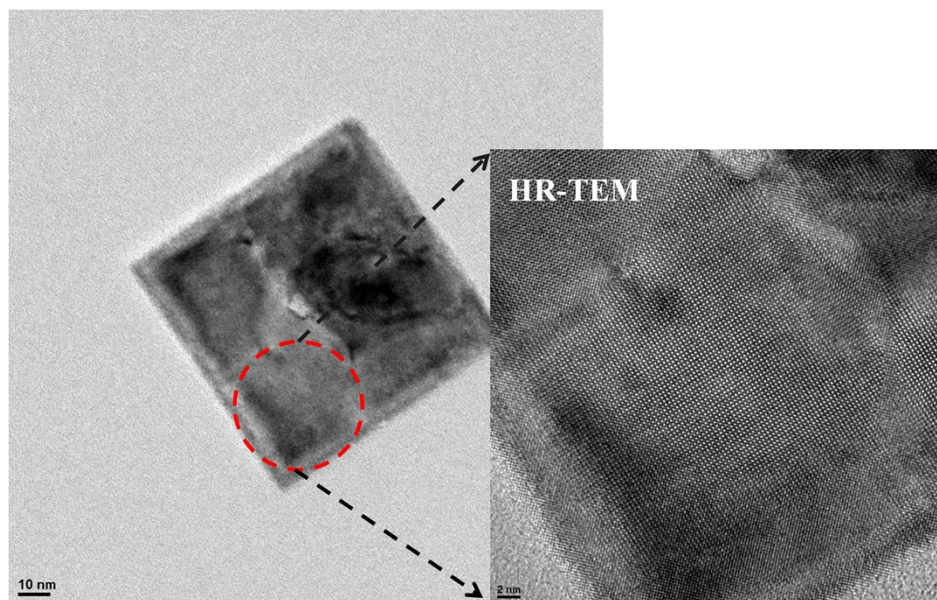


Figure S1: The TEM and HR-TEM images of single FeCr_2Se_4 NCs. Inserted scale bars are 10nm and 2nm, respectively.

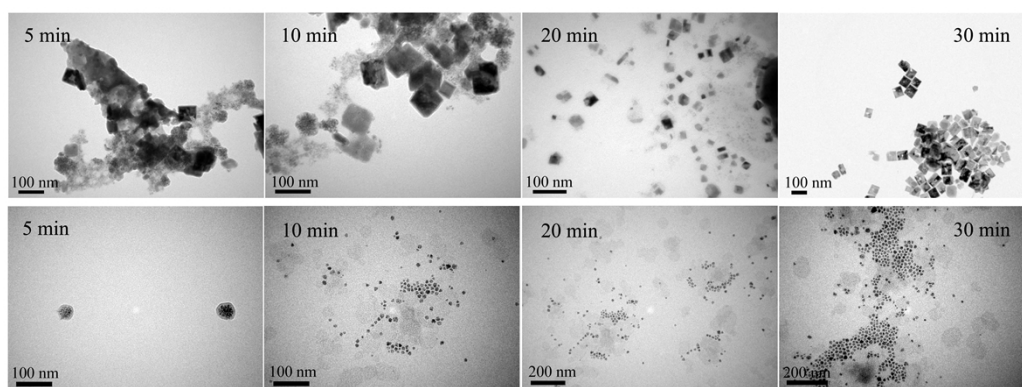


Figure S2: The TEM images of growth process: FeCr_2Se_4 (up) and FeCr_2S_4 (bottom).

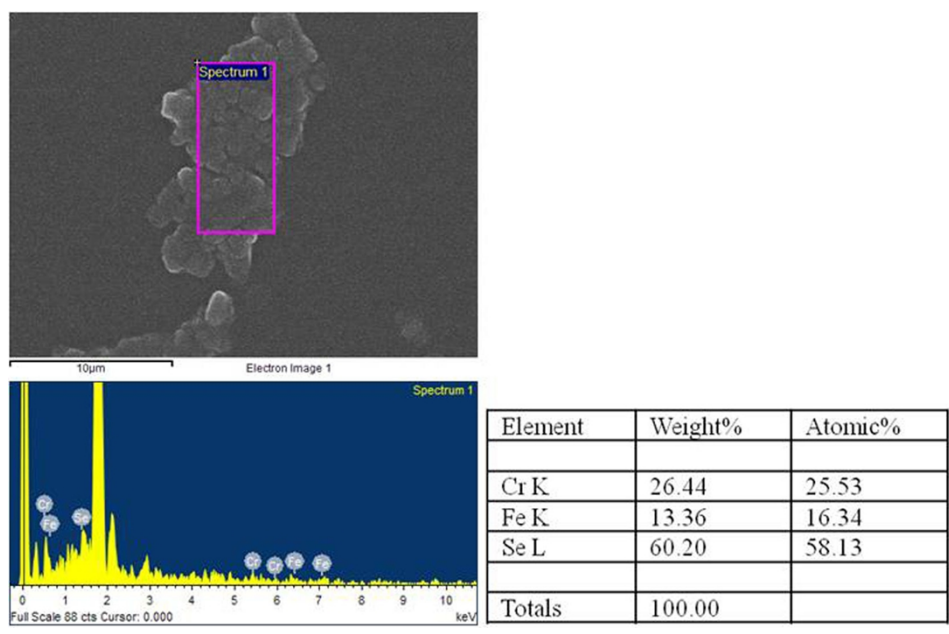


Figure S3: (a) FE-SEM image of OLA-capped NCs. (b) SEM-EDS spectra of FeCr_2Se_4 NCs.

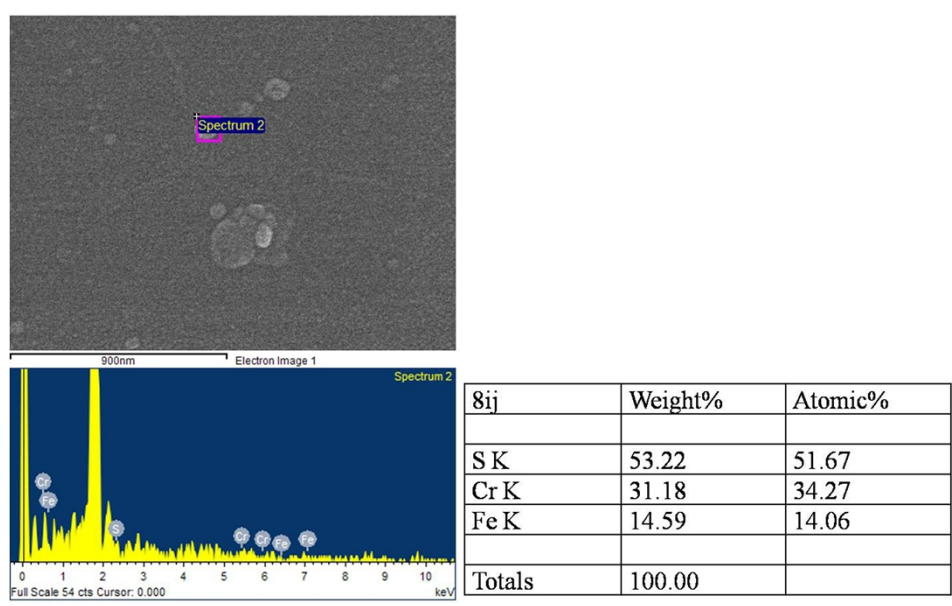


Figure S4: (a) FE-SEM image of OLA-capped NCs. (b) SEM-EDS spectra of FeCr_2S_4 NCs.

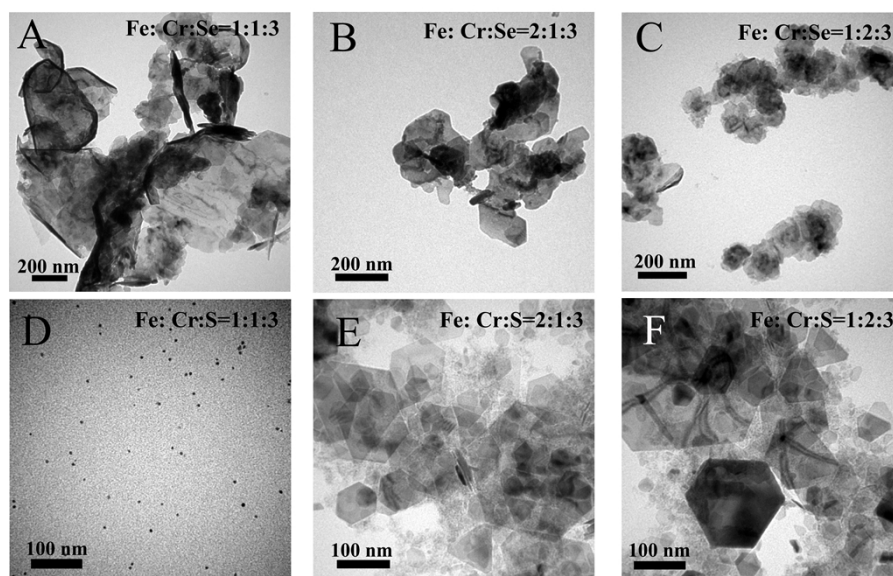


Figure S5. Representative TEM images of the products collected from the reaction with the same condition with different precursor molar ratio used in the synthesis of nanocrystals.

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The TGA data of FeCr_2Se_4 NCs (black) and FeCr_2S_4 NCs (red) were obtained (see Figure S6). For the FeCr_2Se_4 and FeCr_2S_4 NCs, the onset of decomposition occurred at 260 °C and 254 °C, and the decomposition temperature of FeCr_2Se_4 and FeCr_2S_4 NCs were 670 °C and 750 °C with a mass loss of 32.5% and 22.5%, respectively.

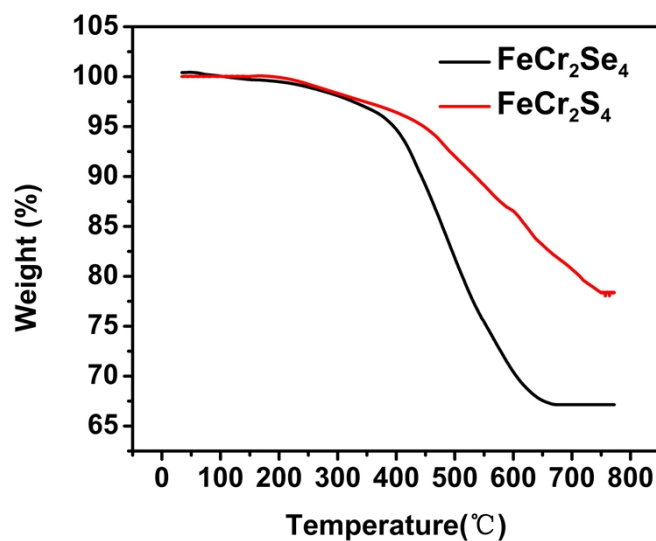


Figure S6: Thermal gravimetric analyses of FeCr_2Se_4 and FeCr_2S_4 NCs.

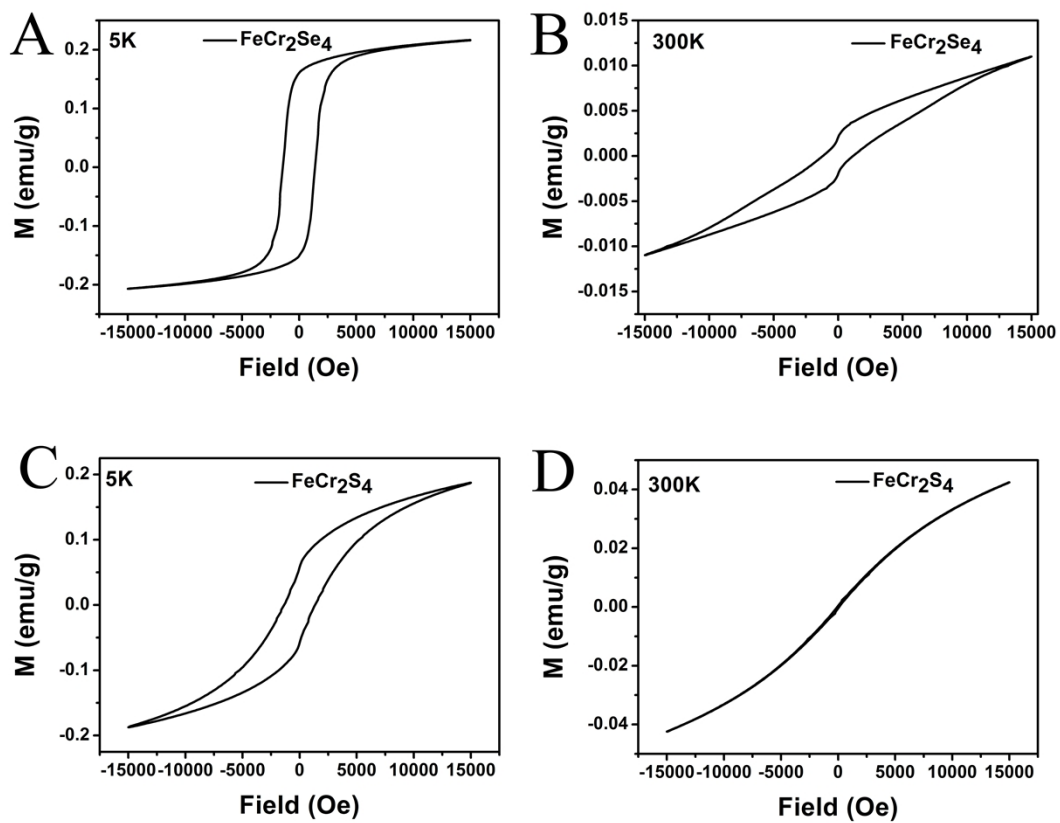


Figure S7 Magnetization (M) as a function of field (H) for FeCr_2Se_4 and FeCr_2S_4 NCs at (A, C) 5 K and (B, D) 300 K

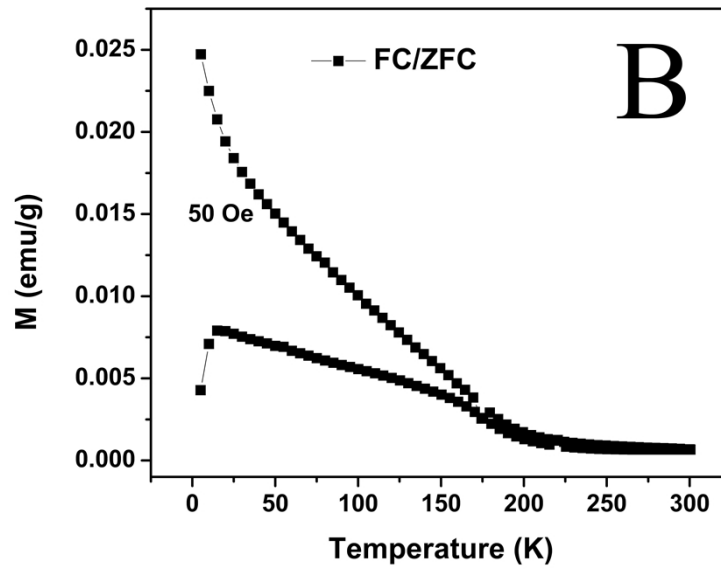
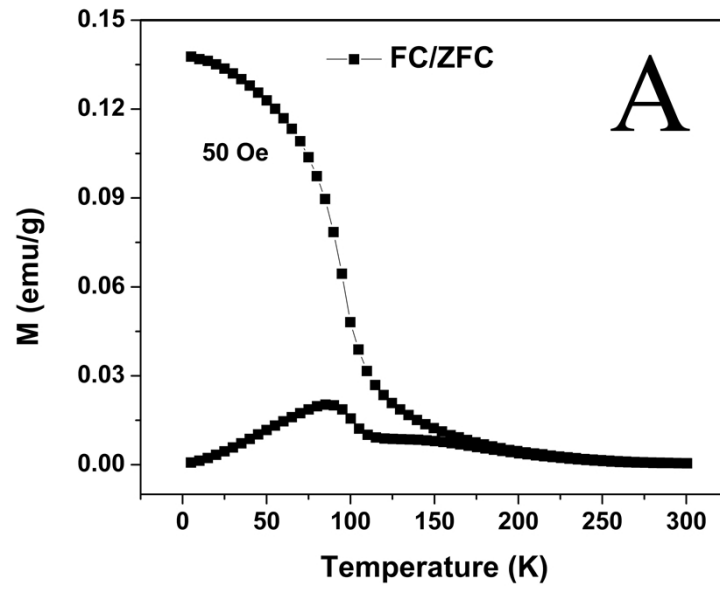


Figure S8: Magnetization (M) as a function of temperature for field-cooled (FC) and zero-field-cooled (ZFC) measurements at 50 Oe for FeCr₂Se₄ (A) and FeCr₂S₄ (B) respectively.

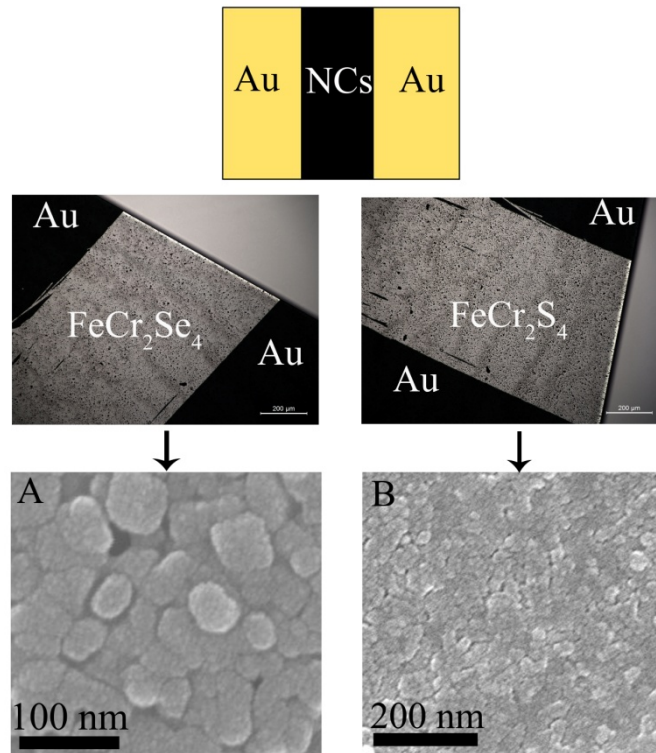


Figure S9: Simple fabrication schematic diagram(upper image) and two point Au electrode pad (insert scale bar is 200 micrometer); and SEM images of FeCr_2Se_4 film (A) and FeCr_2S_4 film (B), respectively.

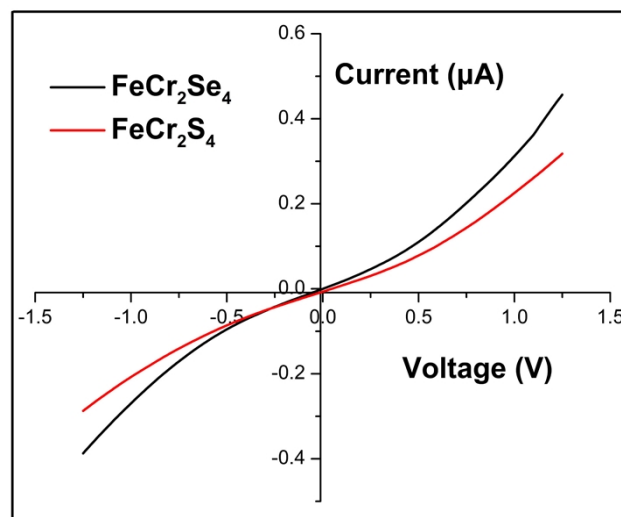


Figure S10: I-V curves of drop-casting films built from NCs, the range of voltage was from $-1.25\text{eV}\sim 1.25\text{eV}$.