# Supporting Information for: Performance Enhancement of Polymer-Based Solar Cells by Induced Phase-Separation with Silica Particles 

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## 1 X-ray photoelectron spectroscopy



Figure S1 XPS data of the PEDOT:PSS coated sapphire substrate. (a) Survey scan (b) Al 2p (c) S 2p (d) Si 2p.

In order to ensure the silica-PEDOT:PSS layer coated the entire substrate, x-ray photoelectron spectroscopy (XPS) was used to probe the atomic composition on the surface, as shown in Fig. S1. Sapphire substrates was used in this experiment to distinguish between the signal from the substrate and the top layer containing PEDOT:PSS and silica. Figure S1 (b) shows the Al 2p signal from the samples coated with PEDOT:PSS with or without silica particles. The absence of the Al 2p peak
is the evidence of complete coverage of the substrate in both cases. The panel (c) shows the S 2p peaks. We found the sulfur signal coming from the surface of PEDOT:PSS is reduced due to the occupation of silica on the substrate. Shown in panel (d) is the Si 2 p signal. As expected, only samples with silica show signals at this energy level. From the XPS results, we conclude the PEDOT:PSS coated the entire substrate surface with or without silica particles. However, the PEDOT:PSS did not completely cover the top of the particle, leaving the surface of silica exposed to the detector. Therefore, we expect those silica surfaces would contact with the subsequently coated active layer.

## 2 Schulz size distribution of spheres

Schulz size distribution is used for many poly-dispersed polymer and colloidal systems. This two-parameter probability distribution function for radii of the spheres is written as:

$$
\begin{equation*}
f(R)=(z+1)^{z+1}\left(\frac{R}{R_{\text {avg }}}\right)^{z} \frac{\exp \left[-(z+1) \frac{R}{R_{\text {avg }}}\right]}{\Gamma(z+1) R_{\text {avg }}} \tag{S1}
\end{equation*}
$$

where $R_{\text {avg }}$ is the number-averaged (arithmetic mean) radius and $z$ is related to polydispersity, $p$, by:

$$
\begin{align*}
& z=\frac{1}{p^{2}}-1  \tag{S2}\\
& p=\frac{\sigma}{R_{\text {avg }}} \tag{S3}
\end{align*}
$$

where $\sigma$ is variance of the distribution.
In a scattering experiment, the scattering intensity is often normalized to the volume of the scatters. The average volume of spheres following Schulz distribution is given as:

$$
\begin{equation*}
\langle V\rangle=\frac{4 \pi}{3}\left\langle R^{3}\right\rangle=\frac{4 \pi}{3} R_{\text {avg }}^{3} \frac{(z+3)(z+2)}{(z+1)^{2}} \tag{S4}
\end{equation*}
$$

where $\left\langle R^{3}\right\rangle$ is the volume-averaged radius.
The scattering intensity from Schulz spheres is:

$$
\begin{equation*}
I(q)=\frac{\phi(\Delta \rho)^{2}}{\langle V\rangle} \int f(R)\left(\frac{4 \pi R^{3}}{3}\right)^{2} \frac{9[\sin (q R)-q R \cos (q R)]^{2}}{(q R)^{6}} d R \tag{S5}
\end{equation*}
$$

Table S1 lists sizes of the silica particles measured by SANS as dilute colloids or coated on silicon wafers with PEDOT:PSS.

Table S1 Sizes of silica nanoparticles from SANS data fitted with Schulz sphere model

| Quantity | Unit | Dilute colloids | Coated on silicon wafers with <br> PEDOT:PSS |
| :---: | :---: | :---: | :---: |
| Volume fraction $\phi$ | - | 0.00591 | 0.0418 |
| Polydispersity $p$ | - | 0.166 | 0.142 |
| $z$ | - | 35.2 | 48.6 |
| $R_{\text {avg }}$ | nm | 62.9 | 65.8 |
| $\left\langle R^{3}\right\rangle$ | nm | 68.1 | 69.8 |

## 3 Guinier analysis on the silica particles

Guinier plots were used to obtain the radii of gyration $\left(R_{g}\right)$ of the silica particles as dilute colloids or coated. The SANS data is ploted as $\ln (I)$ vs. $q^{2}$. In the linear region of the data at small $q$ (Guinier approximation), the $R_{g}$ can be obtained by fitting the data to the following equation:

$$
\begin{equation*}
\ln (I)=\ln \left(I_{0}\right)-\frac{q^{2} R_{g}^{3}}{3} \tag{S6}
\end{equation*}
$$

Notice that $q R_{g}<\sqrt{3}$ to use this approximation. The radius of a sphere can also be obtained by $\sqrt{\frac{5}{3}} R_{g}$.


Figure S2 Guinier plots of silica particles as (a) Dilute colloids. (b) Coated on the silica wafers with PEDOT:PSS.

## 4 Teubner-Strey model and the original fitting parameters

In some conventions, the Teubner-Strey function is written in the following form:

$$
\begin{equation*}
I(q)=\frac{1}{a_{2}^{\prime}+c_{1}^{\prime} q^{2}+c_{2}^{\prime} q^{4}}+b k g \tag{S7}
\end{equation*}
$$

instead of its original version ${ }^{1}$ described in the main text:

$$
\begin{equation*}
I(q)=\frac{\phi_{p}(\Delta \rho)^{2}\left(\frac{8 \pi}{\xi}\right)}{\frac{a_{2}}{c_{2}}+\frac{c_{1}}{c_{2}} q^{2}+q^{4}}+b k g \tag{S8}
\end{equation*}
$$

However, the equations for deriving $d$ (Eqn. 2) and $\xi$ (Eqn. 3) are the same if $a_{2}, c_{1}$ and $c_{2}$ are substituted with $a_{2}^{\prime}, c_{1}^{\prime}$ and $c_{2}^{\prime}$, respectively. The only difference between these two conventions is how the scaling of the equation is incorporated into $c_{2}^{\prime}$. Comparing Eqns. S7 and S8, one can find:

$$
\begin{equation*}
\frac{1}{c_{2}^{\prime}}=\phi_{p}(\Delta \rho)^{2}\left(\frac{8 \pi}{\xi}\right) \tag{S9}
\end{equation*}
$$

or

$$
\begin{equation*}
\phi_{p}(\Delta \rho)^{2}=\left(\frac{\xi}{8 \pi c_{2}^{\prime}}\right)=S F \tag{S10}
\end{equation*}
$$

Here we provide the original Teubner-Strey fitting parameters in the form of Eqn. S7.

Also, The combined model of Schulz spheres and Teubner-Strey used for the SANS data with silica particles is written as:

$$
\begin{align*}
& I(q)=\frac{1}{a_{2}^{\prime}+c_{1}^{\prime} q^{2}+c_{2}^{\prime} q^{4}}+ \\
& \quad \frac{\phi(\Delta \rho)^{2}}{\langle V\rangle} \int f(R)\left(\frac{4 \pi R^{3}}{3}\right)^{2} \frac{9[\sin (q R)-q R \cos (q R)]^{2}}{(q R)^{6}} d R+b k g \tag{S11}
\end{align*}
$$

Tables S2 and S3 list the original fitting parameters for all the SANS data of P3HT:PCBM active layers.

Table S2 Original fitting parameters for samples without the silica particles.

|  |  | PCBM weight $\%$, As-Cast |  |  |  |  |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| Parameter | Unit | $10 \%$ | $20 \%$ | $30 \%$ | $40 \%$ | $50 \%$ |
| $a_{2}^{\prime}$ | $10^{-2} \mathrm{~cm}$ | 8.13 | 2.31 | 1.53 | 1.77 | 0.540 |
| $c_{1}^{\prime}$ | $10^{1} \mathrm{~cm}^{2}$ | -7.03 | -3.40 | -2.38 | -2.37 | -0.557 |
| $c_{2}^{\prime}$ | $10^{4} \mathrm{~cm}^{4}$ | 9.63 | 4.45 | 3.03 | 3.21 | 1.89 |
| $b k g$ | $10^{-1} \mathrm{~cm}^{-1}$ | 5.18 | 3.54 | 3.07 | 5.38 | 3.74 |
| $d$ | $\AA$ | 248 | 267 | 267 | 267 | 340 |
| $\xi$ | $\AA$ | 60.1 | 76.9 | 79.4 | 73.2 | 71.9 |
| $S F$ | $10^{-12} \AA^{-4}$ | 0.248 | 0.688 | 1.04 | 0.907 | 1.51 |
|  |  |  | PCBM weight $\%$, Annealed |  |  |  |
| Parameter | Unit | $10 \%$ | $20 \%$ | $30 \%$ | $40 \%$ | $50 \%$ |
| $a_{2}^{\prime}$ | $10^{-2} \mathrm{~cm}$ | 15.7 | 4.01 | 1.86 | 0.870 | 0.399 |
| $c_{1}^{\prime}$ | $10^{1} \mathrm{~cm} \AA^{2}$ | -16.4 | -6.47 | -3.55 | -1.64 | -0.854 |
| $c_{2}^{\prime}$ | $10^{4} \mathrm{~cm} \AA^{4}$ | 12.7 | 5.69 | 3.21 | 1.98 | 1.79 |
| $b k g$ | $10^{-1} \mathrm{~cm}^{-1}$ | 5.19 | 3.50 | 2.81 | 4.61 | 3.56 |
| $d$ | $\AA$ | 212 | 237 | 245 | 270 | 333 |
| $\xi$ | $\AA$ | 65.4 | 86.0 | 98.0 | 89.9 | 92.5 |
| $S F$ | $10^{-12} \AA^{-4}$ | 0.205 | 0.601 | 1.21 | 1.81 | 2.06 |

Note: $a_{2}^{\prime}, c_{1}^{\prime}, c_{2}^{\prime}$ and $b k g$ are obtained from the fitting results with Eqn. S7. $d, \xi$ and $S F$ are calculated according to Eqns. 2, 3, and S10.

Table S3 Original fitting parameters for samples with the silica particles.

|  |  | PCBM weight $\%$, As-Cast |  |  |  |  |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| Parameter | Unit | $10 \%$ | $20 \%$ | $30 \%$ | $40 \%$ | $50 \%$ |
| $R_{\text {avg }}$ | $\AA$ | 435 | 292 | 243 | 220 | 263 |
| $p$ | - | 0.301 | 0.330 | 0.330 | 0.330 | 0.254 |
| $\Delta \rho$ | $10^{-6} \AA^{-2}$ | 2.68 | 2.68 | 2.68 | 2.68 | 2.68 |
| $\phi$ | $10^{-2}$ | 1.09 | 1.02 | 1.36 | 3.60 | 3.79 |
| $a^{\prime}$ | $10^{-2} \mathrm{~cm}$ | 6.08 | 0.908 | 0.653 | 0.445 | 0.285 |
| $c_{1}^{\prime}$ | $10^{1} \mathrm{~cm} \AA^{2}$ | -2.01 | -0.878 | -1.30 | -0.911 | -0.674 |
| $c_{2}^{\prime}$ | $10^{4} \mathrm{~cm}^{4} \AA^{4}$ | 8.07 | 2.75 | 2.60 | 2.34 | 2.15 |
| $b k g$ | $10^{-1} \mathrm{~cm}^{-1}$ | 4.62 | 9.27 | 5.02 | 4.93 | 7.68 |
| $d$ | $\AA$ | 282 | 328 | 324 | 354 | 389 |
| $\xi$ | $\AA$ | 51.9 | 69.4 | 89.4 | 91.1 | 98.2 |
| $S F$ | $10^{-12} \AA^{-4}$ | 0.256 | 1.01 | 1.37 | 1.55 | 1.82 |


|  |  | PCBM weight \%, Annealed |  |  |  |  |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| Parameter | Unit | $10 \%$ | $20 \%$ | $30 \%$ | $40 \%$ | $50 \%$ |
| $R_{\text {avg }}$ | $\AA$ | 421 | 352 | 242 | 199 | 193 |
| $p$ | - | 0.362 | 0.330 | 0.330 | 0.330 | 0.254 |
| $\Delta \rho$ | $10^{-6} \AA^{-2}$ | 2.68 | 2.68 | 2.68 | 2.68 | 2.68 |
| $\phi$ | $10^{-2}$ | 1.29 | 0.906 | 1.44 | 2.95 | 6.06 |
| $a_{2}^{\prime}$ | $10^{-2} \mathrm{~cm}$ | 13.3 | 1.38 | 0.897 | 0.596 | 0.352 |
| $c_{1}^{\prime}$ | $10^{1} \mathrm{~cm}^{2}$ | -8.88 | -1.23 | -1.79 | -1.58 | -0.869 |
| $c_{2}^{\prime}$ | $10^{4} \mathrm{~cm}^{4}$ | 9.52 | 3.57 | 3.42 | 3.08 | 2.42 |
| $b k g$ | $10^{-1} \mathrm{~cm}^{-1}$ | 5.21 | 10.4 | 6.16 | 6.52 | 8.61 |
| $d$ | $\AA$ | 219 | 315 | 319 | 337 | 375 |
| $\xi$ | $\AA$ | 52.9 | 66.6 | 89.4 | 104 | 99.6 |
| $S F$ | $10^{-12} \AA^{-4}$ | 0.221 | 0.743 | 1.04 | 1.35 | 1.64 |

Note: $R_{\text {avg }}, p, \phi, a_{2}^{\prime}, c_{1}^{\prime}, c_{2}^{\prime}$ and $b k g$ are obtained from the fitting results with Eqn. S11. $\Delta \rho$ is an arbitrary number because the contrast between the silica-PCBM to the matrix is unknown. $d, \xi$ and $S F$ are calculated according to Eqns. 2, 3, and S10.

## 5 Equations for mass conservation and scale factor

For the equation of PCBM mass conservation, assuming the volume of PCBM and P3HT is additive, the weight fraction of PCBM in the solid of the cast solution can be converted to the overall bulk volume fraction
of PCBM:

$$
\begin{equation*}
\phi^{*}=\frac{\frac{w^{*}}{\rho_{P C B M}^{*}}}{\frac{w^{*}}{\rho_{P C B M}^{*}}+\frac{1-w^{*}}{\rho_{P 3 H T}^{*}}} \tag{S12}
\end{equation*}
$$

where
$\phi^{*}$ : Overall bulk volume fraction of PCBM, with respect to the volume of the BHJ film
$w^{*}$ : Weight fraction of PCBM in the solid of the cast solution
$\rho_{\text {PCBM }}^{*}$ : Density of PCBM, ${ }^{2} 1.3 \mathrm{~g} / \mathrm{cm}^{3}$
$\rho_{\text {P3HT }}^{*}$ : Density of P3HT, ${ }^{2} 1.15 \mathrm{~g} / \mathrm{cm}^{3}$
Then, mass conservation of PCBM from the cast solution can be written as:

$$
\begin{equation*}
\phi^{*}=\phi_{p}+\phi_{m} \tag{S13}
\end{equation*}
$$

where
$\phi_{p}$ : Volume fraction of phase-separated PCBM observed in SANS, with respect to the volume of the BHJ film
$\phi_{m}$ : Volume fraction of PCBM dissolved in the matrix, with respect to the volume of the BHJ film

Notice the volume fraction here is based on the total volume of the BHJ. When we discuss the miscibility of PCBM in the matrix, we want the fraction of PCBM in the volume of the matrix. Therefore, we slightly change the notation of $\phi_{m}$ to define $\phi_{m}{ }^{\prime}$, which is the volume fraction of PCBM dissolved in the matrix, with respect to the volume of the matrix. In our two-phase model, the matrix is the space in the BHJ active layer not occupied by the phase-separated PCBM, $\phi_{p}$. Therefore the conversion between $\phi_{m}$ and $\phi_{m}{ }^{\prime}$ is:

$$
\begin{equation*}
\phi_{m}{ }^{\prime}=\frac{\phi_{m}}{1-\phi_{p}} \tag{S14}
\end{equation*}
$$

The denominator means the volume in the BHJ layer excluded by the PCBM phase, which is the volume of the matrix. Combining Eqn. S13 and Eqn. S14, we have

$$
\begin{equation*}
\phi^{*}=\phi_{m}{ }^{\prime}\left(1-\phi_{p}\right)+\phi_{p} \tag{S15}
\end{equation*}
$$

This is the mass balance equation of the system. The subtle difference in the definitions between $\phi_{m}$ and $\phi_{m}{ }^{\prime}$ causes the non-linearity of the mass conservation contour curves on Fig. 8 in the main text.

For the equation of the scale factor, considering the SLD of the matrix as a function of $\phi_{m}{ }^{\prime}$ :

$$
\begin{equation*}
\rho_{\text {matrix }}=\phi_{m}{ }^{\prime} \rho_{P C B M}+\left(1-\phi_{m}{ }^{\prime}\right) \rho_{P 3 H T} \tag{S16}
\end{equation*}
$$

where
$\rho_{\text {matrix }}$ : SLD of the matrix (mixture of P3HT and PCBM)
$\rho_{\text {PCBM }}$ : SLD of PCBM
$\rho_{\text {P3HT }}$ : SLD of P3HT
The contrast term in the system, $\Delta \rho$, is defined as $\left(\rho_{P C B M}-\rho_{\text {matrix }}\right)$. With Eqn. S16 and some simplification, we can rewrite it as:

$$
\begin{equation*}
\Delta \rho=\left(1-\phi_{m}{ }^{\prime}\right)\left(\rho_{P C B M}-\rho_{P 3 H T}\right) \tag{S17}
\end{equation*}
$$

Since $\rho_{P C B M}$ and $\rho_{P 3 H T}$ are known constants, $\Delta \rho$ is a linear transformation of $\phi_{m}{ }^{\prime}$. Therefore, $S F$ can be expressed as $\phi(\Delta \rho)^{2}$, a function of $(\phi, \Delta \rho)$, or a function of $\left(\phi, \phi_{m}{ }^{\prime}\right)$ by using Eqn. S17. We choose the latter form because $\phi_{m}{ }^{\prime}$ indicates the solubility of PCBM in the matrix, a more relevant physical quantity to our interests. Hence,

$$
\begin{equation*}
S F=\phi_{p}\left(1-\phi_{m}{ }^{\prime}\right)^{2}\left(\rho_{P C B M}-\rho_{P 3 H T}\right)^{2} \tag{S18}
\end{equation*}
$$

Now that the mass conservation and scale factor are both explicit functions of $\phi$ and $\phi_{m}{ }^{\prime}$, Fig. 8 can be constructed accordingly.

## 6 Grazing incident x-ray diffraction (GIXRD)



Figure S3 Geometry of the GIXRD setup.
Grazing incident x-ray diffraction (GIXRD) was done with a Rigaku Ultima IV diffractometer by using $\mathrm{CuK} \alpha$ radiation at a fixed incident


Figure S4 GIXRD results of P3HT:PCBM active layers.
angle of 0.6 degree. The receiving side was a pencil detector directly facing the incident $x$-ray beam moving in $2-\theta$ direction. The geometry of the setup is schematically shown in Fig. S3.

The samples were prepared using the same protocol described in the experimental section on the silica wafers coated PEDOT:PSS with or without the silica particles. Each set of as-cast and annealed data was from the same sample before and after annealing. The data is shown in Fig. S4.

We found out the inclusion of silica particles reduced the intensity of
the signal due to the attenuation effect. The attenuation of x-rays can be described by the following equation:

$$
\begin{equation*}
\frac{I}{I_{0}}=\exp \left[-\left(\frac{\mu}{\rho}\right) x\right] \tag{S19}
\end{equation*}
$$

where $\left(\frac{\mu}{\rho}\right)$ is the mass attenuation coefficient, $\rho$ the density of the material, and $x$ the mass thickness. Since $x$ depends on the density of the material, this equation is often used in the following form by substituting $x$ with $\rho t$, where $t$ is the thickness of the sample or x -ray path length:

$$
\begin{equation*}
\frac{I}{I_{0}}=\exp \left[-\left(\frac{\mu}{\rho}\right) \rho t\right] \tag{S20}
\end{equation*}
$$

Due to the shallow incident angle of the x-ray $\left(0.6^{\circ}\right)$, its path length $t$ is essentially parallel to the surface of the film. Therefore, we assume only the x-ray passing through the bottom layer ( $l_{0}$ in Fig. S3) below the top of the silica particles is attenuated by silica particles. We also assume the attenuation effect of the active layers can be neglected compared with that from silica. Since the bottom layer is only partially filled with silica particles with volume fraction, $\phi_{\mathrm{SiO}_{2}}$, Eqn. S 20 is further modified as:

$$
\begin{equation*}
\frac{I}{I_{0}}=\exp \left[-\left(\frac{\mu}{\rho}\right)_{\mathrm{SiO}_{2}} \rho_{\mathrm{SiO}_{2}} t \phi_{\mathrm{SiO}_{2}}\right] \tag{S21}
\end{equation*}
$$

The value of $\left(\frac{\mu}{\rho}\right)_{\mathrm{SiO}_{2}}$ is calculated as $36.4 \mathrm{~cm} \mathrm{~g}^{-2}$ for 8 keV x-ray with the data published by Hubbell and Seltzer, ${ }^{3}$ and $\rho_{\mathrm{SiO}_{2}}$ is $2.32 \mathrm{~g} \mathrm{~cm}^{-1}$ measured by a SANS experiment. Given an estimated $\phi_{\mathrm{SiO}_{2}}$ of 0.2 , we have the attenuation length $L_{e}$ of 0.59 mm , which is the length to reduce $63 \%$ of the intensity. This value is much smaller than the illuminated length of the sample, which we estimated as approximately 20 mm using a phosphor placed on the sample stage. Therefore, we conclude essentially all the x-ray entering the bottom layer was attenuated, so all the diffracted photons by P3HT crystals came from the top layer ( $l_{1}$ ) above the silica particles. In fact, comparing the GIXRD data of samples with or without the silica particles side-by-side, their ratio of intensity is roughly the ratio of $l_{1}$ to the total active layer thickness $\left(l_{0}+l_{1}\right)$. This finding also justifies that only the top region of each sample is measured. However, a quantitative analysis on this system is subjective to many unknown factors, so it is not discussed.

A qualitative analysis of the P3HT (100) peaks shown in Fig. S4 is done with Bragg's law, which gives the d-space of the crystals $(d)$ by:

$$
\begin{equation*}
d=\frac{\lambda}{2 \sin \theta} \tag{S22}
\end{equation*}
$$

where $\lambda$ is $1.54 \AA$.
Scherrer's relation can also be used to estimate the correlation length of the crystalline domains ( $L$ ) given as:

$$
\begin{equation*}
L=\frac{K \lambda}{\beta \cos \theta} \tag{S23}
\end{equation*}
$$

where $K$ is the shape factor chosen as 0.9 here, and $\beta$ is the full-width-half-maximum broadening of the peak.


Figure S5 Analysis of GIXRD data using Bragg's law and Scherrer's relation.
By using Eqns. S22 and S23, we calculated the $d$ and $L$ of each data set as shown in Fig. S5. One can clearly see the silica particles have negligible effect on P3HT crystals in the top layer, because the $d$ 's and $L$ 's
between the treatments with or without silica particles are overlapped except for two cases. The first one is the as-cast $40 \mathrm{Wt} \$.$% PCBM sam-$ ple without silica particles. For some unknown reason, the P3HT approached the state of crystallization as if it had been annealed, and the actual annealing afterwards had no further effect. This is also the same data set showing abnormally low SANS intensity in Fig. 4 (a) in the main text. Therefore, there may be something unexpected happened to this sample, but it does not change the conclusions in this report.

Secondly, with silica particles, the samples with $60 \mathrm{Wt} \$.$% PCBM$ shows a higher Scherrer's correlation length after annealing. However, this is the only exceptional data point, which is insufficient to support that the increased correlation length of P3HT is linked to the rearrangement of PCBM. In fact, we also found the general trend of correlation length is counter-intuitive, as one may expect it to decrease with more PCBM. ${ }^{4}$ However, we currently do not have a satisfactory answer to this finding.

Another observations in the GIXRD results also is the increase in the d-space and the correlation length of the P3HT (100) crystals after annealing, which had been explained in the literature. ${ }^{5,6}$ In summary, due to the attenuation effect of the silica particles, no information on the P3HT crystals near the substrate can be obtained by GIXRD. Also, the silica particles have negligible effect on the P3HT crystals near the top of the active layer based on the current results.

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